

# 4-Methylcinnamaldehyde

<b>Inchi:</b>	InChI=1S/C10H10O/c1-9-4-6-10(7-5-9)3-2-8-11/h2-8H,1H3/b3-2+
<b>InchiKey:</b>	DKOUYOVAEBQFHU-NSCUHMNNSA-N
<b>Formula:</b>	C10H10O
<b>SMILES:</b>	<chem>Cc1ccc(C=CC=O)cc1</chem>
<b>Mol. weight [g/mol]:</b>	146.19
<b>CAS:</b>	71277-10-6

## Physical Properties

Property code	Value	Unit	Source
gf	116.80	kJ/mol	Joback Method
hf	6.97	kJ/mol	Joback Method
hfus	17.80	kJ/mol	Joback Method
hvap	47.47	kJ/mol	Joback Method
log10ws	-2.47		Crippen Method
logp	2.207		Crippen Method
mcvol	125.270	ml/mol	McGowan Method
pc	3318.18	kPa	Joback Method
rinpol	200.50		NIST Webbook
tb	512.68	K	Joback Method
tc	733.50	K	Joback Method
tf	278.32	K	Joback Method
vc	0.484	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	259.86	J/molxK	512.68	Joback Method
cpg	315.28	J/molxK	696.70	Joback Method
cpg	305.68	J/molxK	659.89	Joback Method
cpg	295.38	J/molxK	623.09	Joback Method
cpg	284.34	J/molxK	586.29	Joback Method
cpg	272.52	J/molxK	549.48	Joback Method
cpg	324.23	J/molxK	733.50	Joback Method
dvisc	0.0002293	Paxs	512.68	Joback Method

dvisc	0.0002874	Paxs	473.62	Joback Method
dvisc	0.0003751	Paxs	434.56	Joback Method
dvisc	0.0005159	Paxs	395.50	Joback Method
dvisc	0.0007611	Paxs	356.44	Joback Method
dvisc	0.0012355	Paxs	317.38	Joback Method
dvisc	0.0022976	Paxs	278.32	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C71277106&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C71277106&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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