

5-Bromovaleric acid, 3-methylbut-2-enyl ester

Inchi:	InChI=1S/C10H17BrO2/c1-9(2)6-8-13-10(12)5-3-4-7-11/h6H,3-5,7-8H2,1-2H3
InchiKey:	KMNWYBWUEWICRT-UHFFFAOYSA-N
Formula:	C10H17BrO2
SMILES:	CC(C)=CCOC(=O)CCCCBr
Mol. weight [g/mol]:	249.15

Physical Properties

Property code	Value	Unit	Source
gf	-114.61	kJ/mol	Joback Method
hf	-360.77	kJ/mol	Joback Method
hfus	28.62	kJ/mol	Joback Method
hvap	53.48	kJ/mol	Joback Method
log10ws	-3.16		Crippen Method
logp	3.061		Crippen Method
mcvol	172.400	ml/mol	McGowan Method
pc	2485.07	kPa	Joback Method
tb	574.69	K	Joback Method
tc	770.92	K	Joback Method
tf	315.38	K	Joback Method
vc	0.662	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	393.28	J/molxK	574.69	Joback Method
cpg	406.57	J/molxK	607.39	Joback Method
cpg	419.19	J/molxK	640.10	Joback Method
cpg	431.15	J/molxK	672.80	Joback Method
cpg	442.49	J/molxK	705.51	Joback Method
cpg	453.24	J/molxK	738.21	Joback Method
cpg	463.41	J/molxK	770.92	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U292563&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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