

# Benzeneethanol, «alpha», «alpha»-dimethyl-

<b>Other names:</b>	Phenethyl alcohol, «alpha», «alpha»-dimethyl- «alpha», «alpha»-Dimethylphenethanol «alpha», «alpha»-Dimethylphenethyl alcohol Benzyl­dimethyl­carbinol Dimethylbenzyl­carbinol DMBC 1,1-Dimethyl-2-Phenylethanol 1,1-Dimethylphenylethanol 2-Methyl-1-phenyl-2-propanol Phenyl-tert-butanol «alpha», «alpha»-Dimethyl-«beta»-phenylethyl alcohol «beta»-Phenyl-tert-butyl alcohol 2-Benzyl-2-propanol 2-Hydroxy-2-methyl-1-phenylpropane 2-Methyl-3-phenyl-2-propanol 1,1-Dimethyl-2-phenylethyl alcohol 2-Methyl-1-phenylpropan-2-ol Ethanol, 1,1-dimethyl-2-phenyl- «alpha», «alpha»-Dimethylphenylethyl alcohol NSC 27228
<b>Inchi:</b>	InChI=1S/C10H14O/c1-10(2,11)8-9-6-4-3-5-7-9/h3-7,11H,8H2,1-2H3
<b>InchiKey:</b>	RIWRBSMFKVOJMN-UHFFFAOYSA-N
<b>Formula:</b>	C10H14O
<b>SMILES:</b>	CC(C)(O)Cc1ccccc1
<b>Mol. weight [g/mol]:</b>	150.22
<b>CAS:</b>	100-86-7

## Physical Properties

Property code	Value	Unit	Source
gf	11.75	kJ/mol	Joback Method
hf	-174.18	kJ/mol	Joback Method
hfus	12.37	kJ/mol	Joback Method
h vap	55.51	kJ/mol	Joback Method
log10ws	-2.49		Crippen Method
logp	2.000		Crippen Method
m cvol	133.870	ml/mol	McGowan Method
pc	3318.18	kPa	Joback Method

rropol	1124.60		NIST Webbook
ripol	1781.00		NIST Webbook
tb	543.83	K	Joback Method
tc	747.81	K	Joback Method
tf	292.12	K	Joback Method
vc	0.495	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	315.14	J/mol×K	543.83	Joback Method
cpg	328.48	J/mol×K	577.83	Joback Method
cpg	340.94	J/mol×K	611.82	Joback Method
cpg	352.57	J/mol×K	645.82	Joback Method
cpg	363.42	J/mol×K	679.82	Joback Method
cpg	373.54	J/mol×K	713.82	Joback Method
cpg	382.98	J/mol×K	747.81	Joback Method
dvisc	0.0171729	Paxs	292.12	Joback Method
dvisc	0.0042551	Paxs	334.07	Joback Method
dvisc	0.0014394	Paxs	376.02	Joback Method
dvisc	0.0006053	Paxs	417.97	Joback Method
dvisc	0.0002981	Paxs	459.93	Joback Method
dvisc	0.0001653	Paxs	501.88	Joback Method
dvisc	0.0001003	Paxs	543.83	Joback Method

## Sources

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C100867&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci990307I>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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