

# 2-tert-Butylcyclohexanone

<b>Other names:</b>	2-t-Butylcyclohexanone Cyclohexanone, 2-(1,1-dimethylethyl)-
<b>Inchi:</b>	InChI=1S/C10H18O/c1-10(2,3)8-6-4-5-7-9(8)11/h8H,4-7H2,1-3H3
<b>InchiKey:</b>	ZRYDPLOWJSFQAE-UHFFFAOYSA-N
<b>Formula:</b>	C10H18O
<b>SMILES:</b>	CC(C)(C)C1CCCCC1=O
<b>Mol. weight [g/mol]:</b>	154.25
<b>CAS:</b>	1728-46-7

## Physical Properties

Property code	Value	Unit	Source
gf	-61.98	kJ/mol	Joback Method
hf	-341.86	kJ/mol	Joback Method
hfus	5.59	kJ/mol	Joback Method
hvap	41.23	kJ/mol	Joback Method
log10ws	-2.70		Crippen Method
logp	2.792		Crippen Method
mcvol	142.470	ml/mol	McGowan Method
pc	2715.50	kPa	Joback Method
tb	512.34	K	Joback Method
tc	740.09	K	Joback Method
tf	280.48	K	Joback Method
vc	0.524	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	341.04	J/molxK	512.34	Joback Method
cpg	361.38	J/molxK	550.30	Joback Method
cpg	380.56	J/molxK	588.26	Joback Method
cpg	398.61	J/molxK	626.21	Joback Method
cpg	415.55	J/molxK	664.17	Joback Method
cpg	431.39	J/molxK	702.13	Joback Method
cpg	446.18	J/molxK	740.09	Joback Method

# Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.38300e+01
Coeff. B	-3.83144e+03
Coeff. C	-7.48620e+01
Temperature range (K), min.	357.78
Temperature range (K), max.	524.64

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1728467&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1728467&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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