

# 3-Buten-2-one, 4-(2-chlorophenyl)-

<b>Other names:</b>	3-Buten-2-one, 4-(o-chlorophenyl)- o-Chlorobenzalacetone 4-(2-Chlorophenyl)-but-3-en-2-one (o-Chlorobenzylidene)acetone 4-(2-Chlorophenyl)-3-buten-2-one 2-Chloro-benzalacetone
<b>Inchi:</b>	InChI=1S/C10H9ClO/c1-8(12)6-7-9-4-2-3-5-10(9)11/h2-7H,1H3/b7-6+
<b>InchiKey:</b>	FHDSETHROOWFCQ-VOTSOKGWSA-N
<b>Formula:</b>	C10H9ClO
<b>SMILES:</b>	CC(=O)C=Cc1ccccc1Cl
<b>Mol. weight [g/mol]:</b>	180.63
<b>CAS:</b>	20766-37-4

## Physical Properties

Property code	Value	Unit	Source
gf	75.47	kJ/mol	Joback Method
hf	-35.77	kJ/mol	Joback Method
hfus	21.31	kJ/mol	Joback Method
hvap	51.88	kJ/mol	Joback Method
ie	8.80 ± 0.10	eV	NIST Webbook
ie	8.80	eV	NIST Webbook
ie	8.80	eV	NIST Webbook
log10ws	-3.10		Crippen Method
logp	2.942		Crippen Method
mcvol	137.510	ml/mol	McGowan Method
pc	3142.03	kPa	Joback Method
tb	555.32	K	Joback Method
tc	786.44	K	Joback Method
tf	316.17	K	Joback Method
vc	0.522	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	283.35	J/molxK	555.32	Joback Method
cpg	335.63	J/molxK	747.92	Joback Method
cpg	326.69	J/molxK	709.40	Joback Method
cpg	317.05	J/molxK	670.88	Joback Method
cpg	306.65	J/molxK	632.36	Joback Method
cpg	295.44	J/molxK	593.84	Joback Method
cpg	343.91	J/molxK	786.44	Joback Method
dvisc	0.0002193	Paxs	555.32	Joback Method
dvisc	0.0002751	Paxs	515.46	Joback Method
dvisc	0.0003585	Paxs	475.60	Joback Method
dvisc	0.0004903	Paxs	435.75	Joback Method
dvisc	0.0007142	Paxs	395.89	Joback Method
dvisc	0.0011317	Paxs	356.03	Joback Method
dvisc	0.0020142	Paxs	316.17	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C20766374&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C20766374&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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