

# Benzene, [4-chloro-3-methylbutyl]

<b>Inchi:</b>	InChI=1S/C11H15Cl/c1-10(9-12)7-8-11-5-3-2-4-6-11/h2-6,10H,7-9H2,1H3
<b>InchiKey:</b>	QVCQBXBVPKFMZ-UHFFFAOYSA-N
<b>Formula:</b>	C11H15Cl
<b>SMILES:</b>	CC(CCI)CCc1ccccc1
<b>Mol. weight [g/mol]:</b>	182.69

## Physical Properties

Property code	Value	Unit	Source
gf	139.78	kJ/mol	Joback Method
hf	-54.86	kJ/mol	Joback Method
hfus	18.96	kJ/mol	Joback Method
hvap	46.35	kJ/mol	Joback Method
log10ws	-3.44		Crippen Method
logp	3.494		Crippen Method
mvol	154.330	ml/mol	McGowan Method
pc	2571.50	kPa	Joback Method
tb	514.75	K	Joback Method
tc	726.81	K	Joback Method
tf	255.07	K	Joback Method
vc	0.587	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	330.19	J/mol×K	514.75	Joback Method
cpg	345.85	J/mol×K	550.09	Joback Method
cpg	360.58	J/mol×K	585.44	Joback Method
cpg	374.40	J/mol×K	620.78	Joback Method
cpg	387.37	J/mol×K	656.12	Joback Method
cpg	399.52	J/mol×K	691.46	Joback Method
cpg	410.90	J/mol×K	726.81	Joback Method
dvisc	0.0049594	Paxs	255.07	Joback Method
dvisc	0.0020092	Paxs	298.35	Joback Method
dvisc	0.0010234	Paxs	341.63	Joback Method

dvisc	0.0006066	Paxs	384.91	Joback Method
dvisc	0.0003997	Paxs	428.19	Joback Method
dvisc	0.0002843	Paxs	471.47	Joback Method
dvisc	0.0002142	Paxs	514.75	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R132193&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R132193&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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