

p-Pentylaniline

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|-----------------------------|---|
| Other names: | 4-Pentylaniline Benzenamine, 4-pentyl- p-n-Amylaniline p-n-Pentylaniline |
| Inchi: | InChI=1S/C11H17N/c1-2-3-4-5-10-6-8-11(12)9-7-10/h6-9H,2-5,12H2,1H3 |
| InchiKey: | DGFTWBUZRHAHTH-UHFFFAOYSA-N |
| Formula: | C11H17N |
| SMILES: | CCCCCc1ccc(N)cc1 |
| Mol. weight [g/mol]: | 163.26 |
| CAS: | 33228-44-3 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|---------|----------------|
| gf | 210.97 | kJ/mol | Joback Method |
| hf | -11.52 | kJ/mol | Joback Method |
| hfus | 23.09 | kJ/mol | Joback Method |
| hvap | 53.66 | kJ/mol | Joback Method |
| log10ws | -3.16 | | Crippen Method |
| logp | 3.002 | | Crippen Method |
| mcvol | 152.070 | ml/mol | McGowan Method |
| pc | 2767.17 | kPa | Joback Method |
| tb | 555.27 | K | Joback Method |
| tc | 768.54 | K | Joback Method |
| tf | 335.93 | K | Joback Method |
| vc | 0.573 | m3/kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|--------|---------|-----------------|---------------|
| cpg | 362.66 | J/molxK | 555.27 | Joback Method |
| cpg | 378.10 | J/molxK | 590.82 | Joback Method |
| cpg | 392.67 | J/molxK | 626.36 | Joback Method |
| cpg | 406.38 | J/molxK | 661.91 | Joback Method |
| cpg | 419.28 | J/molxK | 697.45 | Joback Method |

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|-----|--------|---------|--------|---------------|
| cpg | 431.41 | J/mol×K | 733.00 | Joback Method |
| cpg | 442.79 | J/mol×K | 768.54 | Joback Method |

Pressure Dependent Properties

| Property code | Value | Unit | Pressure [kPa] | Source |
|---------------|---------------|------|----------------|--------------|
| tbrp | 418.50 ± 0.50 | K | 2.00 | NIST Webbook |

Correlations

| Information | Value |
|-----------------------------|-------------------------------|
| Property code | pvap |
| Equation | $\ln(P_{vp}) = A + B/(T + C)$ |
| Coeff. A | 1.51922e+01 |
| Coeff. B | -4.62802e+03 |
| Coeff. C | -8.84900e+01 |
| Temperature range (K), min. | 399.00 |
| Temperature range (K), max. | 556.88 |

Sources

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|---|---|
| The Yaws Handbook of Vapor Pressure: | https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci9903071 |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=C33228443&Units=SI |

Legend

| | |
|-------------|---|
| cpg: | Ideal gas heat capacity |
| gf: | Standard Gibbs free energy of formation |

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|-----------------|---|
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| h vap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| pvap: | Vapor pressure |
| tb: | Normal Boiling Point Temperature |
| tbrp: | Boiling point at reduced pressure |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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