

# Dicyclopentylcarbinol

<b>Inchi:</b>	InChI=1S/C11H20O/c12-11(9-5-1-2-6-9)10-7-3-4-8-10/h9-12H,1-8H2
<b>InchiKey:</b>	KJMCBUHSRCZHHX-UHFFFAOYSA-N
<b>Formula:</b>	C11H20O
<b>SMILES:</b>	OC(C1CCCC1)C1CCCC1
<b>Mol. weight [g/mol]:</b>	168.28

## Physical Properties

Property code	Value	Unit	Source
gf	-24.42	kJ/mol	Joback Method
hf	-306.92	kJ/mol	Joback Method
hfus	12.68	kJ/mol	Joback Method
hvap	56.89	kJ/mol	Joback Method
log10ws	-3.11		Crippen Method
logp	2.728		Crippen Method
mcvol	150.000	ml/mol	McGowan Method
pc	3038.96	kPa	Joback Method
rinpol	1408.00		NIST Webbook
rinpol	1408.00		NIST Webbook
tb	573.38	K	Joback Method
tc	781.07	K	Joback Method
tf	281.35	K	Joback Method
vc	0.546	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	406.28	J/molxK	573.38	Joback Method
cpg	490.47	J/molxK	746.46	Joback Method
cpg	475.74	J/molxK	711.84	Joback Method
cpg	460.01	J/molxK	677.23	Joback Method
cpg	443.23	J/molxK	642.61	Joback Method
cpg	425.33	J/molxK	608.00	Joback Method
cpg	504.24	J/molxK	781.07	Joback Method
dvisc	0.0001623	Paxs	573.38	Joback Method

dvisc	0.0002605	Paxs	524.71	Joback Method
dvisc	0.0004606	Paxs	476.04	Joback Method
dvisc	0.0009274	Paxs	427.37	Joback Method
dvisc	0.0022353	Paxs	378.69	Joback Method
dvisc	0.0069844	Paxs	330.02	Joback Method
dvisc	0.0323672	Paxs	281.35	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R120204&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R120204&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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