

4-tert-Butyl-benzophenone

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| Other names: | 4-(t-Butyl)benzophenone Methanone, [4-(1,1-dimethylethyl)phenyl]phenyl- |
| Inchi: | InChI=1S/C17H18O/c1-17(2,3)15-11-9-14(10-12-15)16(18)13-7-5-4-6-8-13/h4-12H,1-3H |
| InchiKey: | DFYJCXSOGSYMAJ-UHFFFAOYSA-N |
| Formula: | C17H18O |
| SMILES: | CC(C)(C)c1ccc(C(=O)c2ccccc2)cc1 |
| Mol. weight [g/mol]: | 238.32 |
| CAS: | 22679-54-5 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|-----------------|----------------------|----------------|
| chl | -9127.00 ± 2.00 | kJ/mol | NIST Webbook |
| gf | 181.37 | kJ/mol | Joback Method |
| hf | -53.95 | kJ/mol | Joback Method |
| hfl | -136.00 | kJ/mol | NIST Webbook |
| hfus | 21.66 | kJ/mol | Joback Method |
| hvap | 64.10 | kJ/mol | Joback Method |
| log10ws | -4.74 | | Crippen Method |
| logp | 4.215 | | Crippen Method |
| mcvol | 204.440 | ml/mol | McGowan Method |
| pc | 2222.89 | kPa | Joback Method |
| tb | 697.34 | K | Joback Method |
| tc | 943.53 | K | Joback Method |
| tf | 399.06 | K | Joback Method |
| vc | 0.766 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|--------|---------|-----------------|---------------|
| cpg | 547.01 | J/mol×K | 697.34 | Joback Method |
| cpg | 564.48 | J/mol×K | 738.37 | Joback Method |
| cpg | 580.50 | J/mol×K | 779.40 | Joback Method |
| cpg | 595.19 | J/mol×K | 820.43 | Joback Method |
| cpg | 608.67 | J/mol×K | 861.47 | Joback Method |

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|-------|-----------|---------|--------|---------------|
| cpg | 621.06 | J/mol×K | 902.50 | Joback Method |
| cpg | 632.46 | J/mol×K | 943.53 | Joback Method |
| dvisc | 0.0015610 | Paxs | 399.06 | Joback Method |
| dvisc | 0.0007988 | Paxs | 448.77 | Joback Method |
| dvisc | 0.0004672 | Paxs | 498.49 | Joback Method |
| dvisc | 0.0003012 | Paxs | 548.20 | Joback Method |
| dvisc | 0.0002089 | Paxs | 597.91 | Joback Method |
| dvisc | 0.0001532 | Paxs | 647.63 | Joback Method |
| dvisc | 0.0001175 | Paxs | 697.34 | Joback Method |

Sources

| | |
|------------------------|---|
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=C22679545&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci9903071 |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |

Legend

| | |
|-----------------|---|
| chl: | Standard liquid enthalpy of combustion |
| cpg: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfl: | Liquid phase enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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