

# 1-Tetradecene, 4,8,12-trimethyl

<b>Inchi:</b>	InChI=1S/C17H34/c1-6-10-16(4)12-9-14-17(5)13-8-11-15(3)7-2/h6,15-17H,1,7-14H2,2-5
<b>InchiKey:</b>	ILDVMZDLEYRGKI-UHFFFAOYSA-N
<b>Formula:</b>	C17H34
<b>SMILES:</b>	C=CCC(C)CCCC(C)CCCC(C)CC
<b>Mol. weight [g/mol]:</b>	238.45

## Physical Properties

Property code	Value	Unit	Source
gf	172.78	kJ/mol	Joback Method
hf	-284.62	kJ/mol	Joback Method
hfus	27.94	kJ/mol	Joback Method
hvap	51.60	kJ/mol	Joback Method
log10ws	-6.07		Crippen Method
logp	6.221		Crippen Method
mvol	246.090	ml/mol	McGowan Method
pc	1290.21	kPa	Joback Method
rinpol	1562.00		NIST Webbook
rinpol	1562.00		NIST Webbook
tb	583.72	K	Joback Method
tc	752.02	K	Joback Method
tf	234.59	K	Joback Method
vc	0.951	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	646.13	J/molxK	583.72	Joback Method
cpg	739.02	J/molxK	723.97	Joback Method
cpg	722.08	J/molxK	695.92	Joback Method
cpg	704.34	J/molxK	667.87	Joback Method
cpg	685.79	J/molxK	639.82	Joback Method
cpg	666.39	J/molxK	611.77	Joback Method
cpg	755.21	J/molxK	752.02	Joback Method
dvisc	0.0001163	Paxs	583.72	Joback Method

dvisc	0.0001687	Paxs	525.53	Joback Method
dvisc	0.0002684	Paxs	467.34	Joback Method
dvisc	0.0004872	Paxs	409.15	Joback Method
dvisc	0.0010779	Paxs	350.97	Joback Method
dvisc	0.0032694	Paxs	292.78	Joback Method
dvisc	0.0171956	Paxs	234.59	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R47063&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R47063&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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