

# 2,4-dichlorobenzyl decyl ether

<b>Inchi:</b>	InChI=1S/C17H26Cl2O/c1-2-3-4-5-6-7-8-9-12-20-14-15-10-11-16(18)13-17(15)19/h10-11
<b>InchiKey:</b>	YFVORVXSIXXPDC-UHFFFAOYSA-N
<b>Formula:</b>	C17H26Cl2O
<b>SMILES:</b>	CCCCCCCCCOCCc1ccc(Cl)cc1Cl
<b>Mol. weight [g/mol]:</b>	317.29

## Physical Properties

Property code	Value	Unit	Source
gf	56.55	kJ/mol	Joback Method
hf	-344.32	kJ/mol	Joback Method
hfus	42.63	kJ/mol	Joback Method
hvap	68.22	kJ/mol	Joback Method
log10ws	-6.99		Crippen Method
logp	6.651		Crippen Method
mcvol	256.980	ml/mol	McGowan Method
pc	1434.80	kPa	Joback Method
rinpol	2171.00		NIST Webbook
rinpol	2179.00		NIST Webbook
rinpol	2177.00		NIST Webbook
tb	722.28	K	Joback Method
tc	918.35	K	Joback Method
tf	414.88	K	Joback Method
vc	0.996	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	693.12	J/molxK	722.28	Joback Method
cpg	709.75	J/molxK	754.96	Joback Method
cpg	725.45	J/molxK	787.64	Joback Method
cpg	740.26	J/molxK	820.32	Joback Method
cpg	754.19	J/molxK	852.99	Joback Method
cpg	767.27	J/molxK	885.67	Joback Method
cpg	779.54	J/molxK	918.35	Joback Method

dvisc	0.0009350	Paxs	414.88	Joback Method
dvisc	0.0005074	Paxs	466.11	Joback Method
dvisc	0.0003108	Paxs	517.35	Joback Method
dvisc	0.0002080	Paxs	568.58	Joback Method
dvisc	0.0001487	Paxs	619.81	Joback Method
dvisc	0.0001119	Paxs	671.05	Joback Method
dvisc	0.0000877	Paxs	722.28	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R32254&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R32254&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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