

# 3,4,4a,9a-Tetrahydrofluorene

<b>Other names:</b>	1,4,4a,9a-tetrahydrofluorene
<b>Inchi:</b>	InChI=1S/C13H14/c1-3-7-12-10(5-1)9-11-6-2-4-8-13(11)12/h1-3,5-7,11,13H,4,8-9H2
<b>InchiKey:</b>	FQADUXRTNGLZSX-UHFFFAOYSA-N
<b>Formula:</b>	C13H14
<b>SMILES:</b>	C1=CC2Cc3ccccc3C2CC1
<b>Mol. weight [g/mol]:</b>	170.25
<b>CAS:</b>	52652-40-1

## Physical Properties

Property code	Value	Unit	Source
gf	300.72	kJ/mol	Joback Method
hf	110.63	kJ/mol	Joback Method
hfus	18.47	kJ/mol	Joback Method
hvap	47.76	kJ/mol	Joback Method
log10ws	-3.70		Crippen Method
logp	3.292		Crippen Method
mcvol	144.250	ml/mol	McGowan Method
pc	2989.32	kPa	Joback Method
tb	545.41	K	Joback Method
tc	785.78	K	Joback Method
tf	308.33	K	Joback Method
vc	0.547	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	348.02	J/molxK	545.41	Joback Method
cpg	367.14	J/molxK	585.47	Joback Method
cpg	384.74	J/molxK	625.53	Joback Method
cpg	400.96	J/molxK	665.60	Joback Method
cpg	415.90	J/molxK	705.66	Joback Method
cpg	429.69	J/molxK	745.72	Joback Method
cpg	442.46	J/molxK	785.78	Joback Method
dvisc	0.0016778	Paxs	308.33	Joback Method

dvisc	0.0013438	Paxs	347.84	Joback Method
dvisc	0.0011261	Paxs	387.36	Joback Method
dvisc	0.0009751	Paxs	426.87	Joback Method
dvisc	0.0008651	Paxs	466.38	Joback Method
dvisc	0.0007821	Paxs	505.90	Joback Method
dvisc	0.0007174	Paxs	545.41	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C52652401&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C52652401&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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