

# 4-(2,6,6-Trimethylcyclohexa-1,3-dienyl)but-3-en-2-

<b>Other names:</b>	3,4-Dehydro-«beta»-ionone Dehydro-«beta»-ionone 3,4-didehydro-«beta»-ionone dehydroionone
<b>Inchi:</b>	InChI=1S/C13H18O/c1-10-6-5-9-13(3,4)12(10)8-7-11(2)14/h5-8H,9H2,1-4H3/b8-7+
<b>InchiKey:</b>	UWWCASOGCPOGJP-BQYQJAHWSA-N
<b>Formula:</b>	C13H18O
<b>SMILES:</b>	CC(=O)C=CC1=C(C)C=CCC1(C)C
<b>Mol. weight [g/mol]:</b>	190.28
<b>CAS:</b>	1203-08-3

## Physical Properties

Property code	Value	Unit	Source
gf	69.50	kJ/mol	Joback Method
hf	-144.83	kJ/mol	Joback Method
hfus	18.43	kJ/mol	Joback Method
hvap	52.42	kJ/mol	Joback Method
log10ws	-3.76		Crippen Method
logp	3.434		Crippen Method
mcvol	171.840	ml/mol	McGowan Method
pc	2358.78	kPa	Joback Method
rinpol	1485.00		NIST Webbook
rinpol	1485.00		NIST Webbook
rinpol	1485.30		NIST Webbook
rinpol	1460.00		NIST Webbook
rinpol	1487.00		NIST Webbook
rinpol	1485.30		NIST Webbook
tb	582.94	K	Joback Method
tc	804.13	K	Joback Method
tf	338.96	K	Joback Method
vc	0.652	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	417.12	J/mol×K	582.94	Joback Method
cpg	433.84	J/mol×K	619.80	Joback Method
cpg	449.57	J/mol×K	656.67	Joback Method
cpg	464.42	J/mol×K	693.53	Joback Method
cpg	478.54	J/mol×K	730.40	Joback Method
cpg	492.06	J/mol×K	767.26	Joback Method
cpg	505.11	J/mol×K	804.13	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1203083&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1203083&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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