

# 1-Tridecanamine

<b>Other names:</b>	1-Aminotridecane Monotridecylamine Tridecylamine n-Tridecylamine
<b>Inchi:</b>	InChI=1S/C13H29N/c1-2-3-4-5-6-7-8-9-10-11-12-13-14/h2-14H2,1H3
<b>InchiKey:</b>	ABVVEAHYODGCLZ-UHFFFAOYSA-N
<b>Formula:</b>	C13H29N
<b>SMILES:</b>	CCCCCCCCCCCCCN
<b>Mol. weight [g/mol]:</b>	199.38
<b>CAS:</b>	2869-34-3

## Physical Properties

Property code	Value	Unit	Source
gf	125.03	kJ/mol	Joback Method
hf	-277.86	kJ/mol	Joback Method
hfus	34.62	kJ/mol	Joback Method
hvap	55.17	kJ/mol	Joback Method
log10ws	-4.70		Crippen Method
logp	4.256		Crippen Method
mcvol	204.010	ml/mol	McGowan Method
pc	1723.16	kPa	Joback Method
tb	538.15 ± 6.00	K	NIST Webbook
tb	548.00	K	NIST Webbook
tb	538.20	K	NIST Webbook
tb	703.15 ± 4.00	K	NIST Webbook
tc	739.84	K	Joback Method
tf	300.65 ± 2.00	K	NIST Webbook
tf	300.15 ± 2.00	K	NIST Webbook
vc	0.792	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	531.05	J/molxK	569.37	Joback Method

cpg	548.39	J/mol×K	597.78	Joback Method
cpg	565.01	J/mol×K	626.19	Joback Method
cpg	580.92	J/mol×K	654.60	Joback Method
cpg	596.14	J/mol×K	683.02	Joback Method
cpg	610.71	J/mol×K	711.43	Joback Method
cpg	624.63	J/mol×K	739.84	Joback Method
hvapt	76.80	kJ/mol	652.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.61780e+01
Coeff. B	-5.23738e+03
Coeff. C	-9.49260e+01
Temperature range (K), min.	424.52
Temperature range (K), max.	576.90

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2869343&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2869343&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions

<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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