

# 2-[2-[2-[2-(2-Methoxyethoxy)ethoxy]ethoxy]ethoxy]ethoxy

Other names:  
**acetate**

Pentaethylene glycol monomethyl ether, acetate

3,6,9,12,15-Heptaoxahexadec-1-yl acetate

**Inchi:** InChI=1S/C13H26O7/c1-13(14)20-12-11-19-10-9-18-8-7-17-6-5-16-4-3-15-2/h3-12H2,1-2

**InchiKey:** CJUOGNAQUZKQPS-UHFFFAOYSA-N

**Formula:** C13H26O7

**SMILES:** COCCOCCOCCOCCOCCOC(C)=O

**Mol. weight [g/mol]:** 294.34

## Physical Properties

Property code	Value	Unit	Source
gf	-700.34	kJ/mol	Joback Method
hf	-1217.55	kJ/mol	Joback Method
hfus	38.15	kJ/mol	Joback Method
hvap	65.74	kJ/mol	Joback Method
log10ws	0.44		Crippen Method
logp	0.262		Crippen Method
mcvol	230.820	ml/mol	McGowan Method
pc	1593.62	kPa	Joback Method
rinpol	1938.40		NIST Webbook
rinpol	1938.40		NIST Webbook
tb	685.23	K	Joback Method
tc	855.68	K	Joback Method
tf	419.58	K	Joback Method
vc	0.877	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	659.38	J/molxK	685.23	Joback Method
cpg	674.95	J/molxK	713.64	Joback Method
cpg	689.85	J/molxK	742.05	Joback Method
cpg	704.06	J/molxK	770.45	Joback Method
cpg	717.55	J/molxK	798.86	Joback Method
cpg	730.28	J/molxK	827.27	Joback Method

cpg	742.24	J/mol×K	855.68	Joback Method
dvisc	0.0004899	Paxs	419.58	Joback Method
dvisc	0.0002766	Paxs	463.86	Joback Method
dvisc	0.0001725	Paxs	508.13	Joback Method
dvisc	0.0001160	Paxs	552.40	Joback Method
dvisc	0.0000828	Paxs	596.68	Joback Method
dvisc	0.0000619	Paxs	640.96	Joback Method
dvisc	0.0000480	Paxs	685.23	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U351913&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U351913&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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