

4-methyl-2-phenyl-2-hexenal

Inchi:	InChI=1S/C13H16O/c1-3-11(2)9-13(10-14)12-7-5-4-6-8-12/h4-11H,3H2,1-2H3/b13-9-
InchiKey:	IOIWDGZFMUCYJR-LCYFTJDESA-N
Formula:	C13H16O
SMILES:	CCC(C)C=C(C=O)c1ccccc1
Mol. weight [g/mol]:	188.27

Physical Properties

Property code	Value	Unit	Source
gf	140.70	kJ/mol	Joback Method
hf	-58.55	kJ/mol	Joback Method
hfus	21.13	kJ/mol	Joback Method
hvap	53.18	kJ/mol	Joback Method
log10ws	-3.43		Crippen Method
logp	3.315		Crippen Method
mcvol	167.540	ml/mol	McGowan Method
pc	2510.03	kPa	Joback Method
ripol	2000.00		NIST Webbook
ripol	2000.00		NIST Webbook
tb	575.78	K	Joback Method
tc	792.99	K	Joback Method
tf	270.65	K	Joback Method
vc	0.647	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	398.02	J/mol×K	575.78	Joback Method
cpg	414.02	J/mol×K	611.98	Joback Method
cpg	428.96	J/mol×K	648.18	Joback Method
cpg	442.92	J/mol×K	684.39	Joback Method
cpg	455.95	J/mol×K	720.59	Joback Method
cpg	468.11	J/mol×K	756.79	Joback Method
cpg	479.46	J/mol×K	792.99	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R308182&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
ripol:	Polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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