

# 3-methylbutyl hexadecanoate

<b>Other names:</b>	isopentyl hexadecanoate 3-methylbutyl palmitate
<b>Inchi:</b>	InChI=1S/C21H42O2/c1-4-5-6-7-8-9-10-11-12-13-14-15-16-17-21(22)23-19-18-20(2)3/h2
<b>InchiKey:</b>	NXBIDOXPIIWDMX-UHFFFAOYSA-N
<b>Formula:</b>	C21H42O2
<b>SMILES:</b>	CCCCCCCCCCCCCCCC(=O)OCCC(C)C
<b>Mol. weight [g/mol]:</b>	326.56
<b>CAS:</b>	81974-61-0

## Physical Properties

Property code	Value	Unit	Source
gf	-110.42	kJ/mol	Joback Method
hf	-726.85	kJ/mol	Joback Method
hfus	49.41	kJ/mol	Joback Method
hvap	71.11	kJ/mol	Joback Method
log10ws	-7.23		Crippen Method
logp	7.057		Crippen Method
mcvol	314.190	ml/mol	McGowan Method
pc	988.88	kPa	Joback Method
rinpol	2246.00		NIST Webbook
rinpol	2246.00		NIST Webbook
rinpol	2260.00		NIST Webbook
rinpol	2260.00		NIST Webbook
ripol	2479.00		NIST Webbook
ripol	2479.00		NIST Webbook
tb	755.73	K	Joback Method
tc	930.28	K	Joback Method
tf	383.59	K	Joback Method
vc	1.230	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	969.75	J/molxK	755.73	Joback Method

cpg	1061.34	J/mol×K	901.19	Joback Method
cpg	1044.89	J/mol×K	872.09	Joback Method
cpg	1027.53	J/mol×K	843.00	Joback Method
cpg	1009.24	J/mol×K	813.91	Joback Method
cpg	989.99	J/mol×K	784.82	Joback Method
cpg	1076.91	J/mol×K	930.28	Joback Method
dvisc	0.0000590	Paxs	755.73	Joback Method
dvisc	0.0000811	Paxs	693.71	Joback Method
dvisc	0.0001187	Paxs	631.68	Joback Method
dvisc	0.0001889	Paxs	569.66	Joback Method
dvisc	0.0003366	Paxs	507.64	Joback Method
dvisc	0.0007045	Paxs	445.61	Joback Method
dvisc	0.0018724	Paxs	383.59	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C81974610&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C81974610&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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