

# 1,2-Cyclohexanedicarboxylic acid, 3,5-dimethylcyclohexyl pentyl ester

Inchi:	InChI=1S/C21H36O4/c1-4-5-8-11-24-20(22)18-9-6-7-10-19(18)21(23)25-17-13-15(2)12-
InchiKey:	CNSKQIPMGXCKKR-UHFFFAOYSA-N
Formula:	C21H36O4
SMILES:	CCCCCOC(=O)C1CCCCC1C(=O)OC1CC(C)CC(C)C1
Mol. weight [g/mol]:	352.51

## Physical Properties

Property code	Value	Unit	Source
gf	-316.13	kJ/mol	Joback Method
hf	-918.75	kJ/mol	Joback Method
hfus	42.60	kJ/mol	Joback Method
hvap	80.58	kJ/mol	Joback Method
log10ws	-5.27		Crippen Method
logp	4.894		Crippen Method
mcvol	299.910	ml/mol	McGowan Method
pc	1247.73	kPa	Joback Method
rinpol	2369.00		NIST Webbook
rinpol	2369.00		NIST Webbook
tb	857.55	K	Joback Method
tc	1070.02	K	Joback Method
tf	472.79	K	Joback Method
vc	1.123	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1035.08	J/molxK	857.55	Joback Method
cpg	1055.95	J/molxK	892.96	Joback Method
cpg	1074.98	J/molxK	928.37	Joback Method
cpg	1092.18	J/molxK	963.79	Joback Method
cpg	1107.58	J/molxK	999.20	Joback Method
cpg	1121.17	J/molxK	1034.61	Joback Method
cpg	1132.97	J/molxK	1070.02	Joback Method
dvisc	0.0011802	Paxs	472.79	Joback Method

dvisc	0.0006229	Paxs	536.92	Joback Method
dvisc	0.0003768	Paxs	601.04	Joback Method
dvisc	0.0002511	Paxs	665.17	Joback Method
dvisc	0.0001797	Paxs	729.30	Joback Method
dvisc	0.0001358	Paxs	793.42	Joback Method
dvisc	0.0001070	Paxs	857.55	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U339848&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U339848&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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