

# 1,2-Cyclohexanedicarboxylic acid, isobutyl 2-isopropylphenyl ester

<b>Inchi:</b>	InChI=1S/C21H30O4/c1-14(2)13-24-20(22)17-10-5-6-11-18(17)21(23)25-19-12-8-7-9-16
<b>InchiKey:</b>	ZNBBEVRBWHGRPU-UHFFFAOYSA-N
<b>Formula:</b>	C21H30O4
<b>SMILES:</b>	CC(C)COC(=O)C1CCCCC1C(=O)Oc1ccccc1C(C)C
<b>Mol. weight [g/mol]:</b>	346.46

## Physical Properties

Property code	Value	Unit	Source
gf	-227.26	kJ/mol	Joback Method
hf	-717.89	kJ/mol	Joback Method
hfus	35.23	kJ/mol	Joback Method
hvap	82.93	kJ/mol	Joback Method
log10ws	-5.19		Crippen Method
logp	4.721		Crippen Method
mvol	287.010	ml/mol	McGowan Method
pc	1436.98	kPa	Joback Method
rinpol	2341.00		NIST Webbook
rinpol	2341.00		NIST Webbook
tb	878.12	K	Joback Method
tc	1101.11	K	Joback Method
tf	482.83	K	Joback Method
vc	1.071	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	937.49	J/molxK	878.12	Joback Method
cpg	955.09	J/molxK	915.29	Joback Method
cpg	971.02	J/molxK	952.45	Joback Method
cpg	985.32	J/molxK	989.62	Joback Method
cpg	998.00	J/molxK	1026.78	Joback Method
cpg	1009.11	J/molxK	1063.95	Joback Method
cpg	1018.67	J/molxK	1101.11	Joback Method
dvisc	0.0008474	Paxs	482.83	Joback Method

dvisc	0.0004060	Paxs	548.71	Joback Method
dvisc	0.0002278	Paxs	614.59	Joback Method
dvisc	0.0001429	Paxs	680.47	Joback Method
dvisc	0.0000974	Paxs	746.36	Joback Method
dvisc	0.0000706	Paxs	812.24	Joback Method
dvisc	0.0000537	Paxs	878.12	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U339699&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U339699&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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