

# Glutaric acid, hept-2-yl 2-isopropoxyphenyl ester

Inchi:	InChI=1S/C21H32O5/c1-5-6-7-11-17(4)25-20(22)14-10-15-21(23)26-19-13-9-8-12-18(19)
InchiKey:	OSHAQARDAROVBS-UHFFFAOYSA-N
Formula:	C21H32O5
SMILES:	CCCCC(C)OC(=O)CCCC(=O)Oc1ccccc1OC(C)C
Mol. weight [g/mol]:	364.48

## Physical Properties

Property code	Value	Unit	Source
gf	-349.00	kJ/mol	Joback Method
hf	-884.09	kJ/mol	Joback Method
hfus	43.51	kJ/mol	Joback Method
hvap	85.22	kJ/mol	Joback Method
log10ws	-6.01		Crippen Method
logp	5.062		Crippen Method
mvol	303.740	ml/mol	McGowan Method
pc	1251.26	kPa	Joback Method
rinpol	2418.00		NIST Webbook
rinpol	2418.00		NIST Webbook
tb	885.66	K	Joback Method
tc	1091.46	K	Joback Method
tf	501.92	K	Joback Method
vc	1.157	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	967.72	J/molxK	885.66	Joback Method
cpg	1034.38	J/molxK	1057.16	Joback Method
cpg	1023.65	J/molxK	1022.86	Joback Method
cpg	1011.63	J/molxK	988.56	Joback Method
cpg	998.32	J/molxK	954.26	Joback Method
cpg	983.68	J/molxK	919.96	Joback Method
cpg	1043.83	J/molxK	1091.46	Joback Method
dvisc	0.0000298	Paxs	885.66	Joback Method

dvisc	0.0000394	Paxs	821.70	Joback Method
dvisc	0.0000548	Paxs	757.75	Joback Method
dvisc	0.0000809	Paxs	693.79	Joback Method
dvisc	0.0001293	Paxs	629.83	Joback Method
dvisc	0.0002299	Paxs	565.88	Joback Method
dvisc	0.0004729	Paxs	501.92	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U391872&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U391872&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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