

# Fumaric acid, 3,5-dimethylphenyl 2-ethylhexyl ester

Inchi:	InChI=1S/C20H28O4/c1-5-7-8-17(6-2)14-23-19(21)9-10-20(22)24-18-12-15(3)11-16(4)13
InchiKey:	WSDNHSPDPXRVKI-MDZDMXLPSA-N
Formula:	C20H28O4
SMILES:	CCCCC(CC)COC(=O)C=CC(=O)Oc1cc(C)cc(C)c1
Mol. weight [g/mol]:	332.43

## Physical Properties

Property code	Value	Unit	Source
gf	-179.39	kJ/mol	Joback Method
hf	-620.20	kJ/mol	Joback Method
hfus	43.07	kJ/mol	Joback Method
hvap	81.60	kJ/mol	Joback Method
log10ws	-5.40		Crippen Method
logp	4.525		Crippen Method
mcvol	279.480	ml/mol	McGowan Method
pc	1387.11	kPa	Joback Method
rinsol	2414.00		NIST Webbook
tb	849.94	K	Joback Method
tc	1056.88	K	Joback Method
tf	490.86	K	Joback Method
vc	1.069	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	851.04	J/molxK	849.94	Joback Method
cpg	866.76	J/molxK	884.43	Joback Method
cpg	881.38	J/molxK	918.92	Joback Method
cpg	894.91	J/molxK	953.41	Joback Method
cpg	907.40	J/molxK	987.90	Joback Method
cpg	918.87	J/molxK	1022.39	Joback Method
cpg	929.34	J/molxK	1056.88	Joback Method
dvisc	0.0005271	Paxs	490.86	Joback Method
dvisc	0.0002789	Paxs	550.71	Joback Method

dvisc	0.0001672	Paxs	610.55	Joback Method
dvisc	0.0001098	Paxs	670.40	Joback Method
dvisc	0.0000773	Paxs	730.25	Joback Method
dvisc	0.0000573	Paxs	790.09	Joback Method
dvisc	0.0000444	Paxs	849.94	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U405740&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U405740&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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