

# 4-Pentenoic acid, 2-methyl-, tetradecyl ester

**Inchi:** InChI=1S/C20H38O2/c1-4-6-7-8-9-10-11-12-13-14-15-16-18-22-20(21)19(3)17-5-2/h5,19  
**InchiKey:** AITVUERWEKFFAB-UHFFFAOYSA-N  
**Formula:** C20H38O2  
**SMILES:** C=CCC(C)C(=O)OCCCCCCCCCCCCCCC  
**Mol. weight [g/mol]:** 310.51

## Physical Properties

Property code	Value	Unit	Source
gf	-31.00	kJ/mol	Joback Method
hf	-580.78	kJ/mol	Joback Method
hfus	45.54	kJ/mol	Joback Method
hvap	68.21	kJ/mol	Joback Method
log10ws	-6.67		Crippen Method
logp	6.443		Crippen Method
mcvol	295.800	ml/mol	McGowan Method
pc	1080.64	kPa	Joback Method
rinpol	2110.00		NIST Webbook
rinpol	2110.00		NIST Webbook
tb	729.53	K	Joback Method
tc	902.98	K	Joback Method
tf	370.56	K	Joback Method
vc	1.155	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	883.57	J/molxK	729.53	Joback Method
cpg	902.88	J/molxK	758.44	Joback Method
cpg	921.27	J/molxK	787.35	Joback Method
cpg	938.76	J/molxK	816.26	Joback Method
cpg	955.38	J/molxK	845.16	Joback Method
cpg	971.16	J/molxK	874.07	Joback Method
cpg	986.12	J/molxK	902.98	Joback Method
dvisc	0.0020748	Paxs	370.56	Joback Method

dvisc	0.0008018	Paxs	430.39	Joback Method
dvisc	0.0003908	Paxs	490.22	Joback Method
dvisc	0.0002227	Paxs	550.04	Joback Method
dvisc	0.0001417	Paxs	609.87	Joback Method
dvisc	0.0000977	Paxs	669.70	Joback Method
dvisc	0.0000717	Paxs	729.53	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U406117&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U406117&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/80-498-1/4-Pentenoic-acid-2-methyl-tetradecyl-ester.pdf>

Generated by Cheméo on 2024-04-19 19:21:08.183701846 +0000 UTC m=+15843717.104279160.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.