

# Diglycolic acid, 3,7-dimethyloctyl isohexyl ester

Inchi:	InChI=1S/C20H38O5/c1-16(2)8-6-10-18(5)11-13-25-20(22)15-23-14-19(21)24-12-7-9-17
InchiKey:	ZIYOCDSIMZRRNO-UHFFFAOYSA-N
Formula:	C20H38O5
SMILES:	CC(C)CCCOC(=O)COCC(=O)OCCC(C)CCCC(C)C
Mol. weight [g/mol]:	358.51

## Physical Properties

Property code	Value	Unit	Source
gf	-462.64	kJ/mol	Joback Method
hf	-1093.79	kJ/mol	Joback Method
hfus	43.75	kJ/mol	Joback Method
hvap	79.67	kJ/mol	Joback Method
log10ws	-4.28		Crippen Method
logp	4.378		Crippen Method
mcvol	313.410	ml/mol	McGowan Method
pc	1082.78	kPa	Joback Method
rinsol	2830.00		NIST Webbook
tb	830.68	K	Joback Method
tc	1019.79	K	Joback Method
tf	436.71	K	Joback Method
vc	1.204	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1002.28	J/molxK	830.68	Joback Method
cpg	1020.47	J/molxK	862.20	Joback Method
cpg	1037.49	J/molxK	893.72	Joback Method
cpg	1053.34	J/molxK	925.24	Joback Method
cpg	1068.02	J/molxK	956.75	Joback Method
cpg	1081.56	J/molxK	988.27	Joback Method
cpg	1093.95	J/molxK	1019.79	Joback Method
dvisc	0.0009558	Paxs	436.71	Joback Method
dvisc	0.0003706	Paxs	502.37	Joback Method

dvisc	0.0001789	Paxs	568.03	Joback Method
dvisc	0.0001004	Paxs	633.69	Joback Method
dvisc	0.0000628	Paxs	699.36	Joback Method
dvisc	0.0000426	Paxs	765.02	Joback Method
dvisc	0.0000307	Paxs	830.68	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U382147&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U382147&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/80-531-3/Diglycolic-acid-3-7-dimethyloctyl-isohehexyl-ester.pdf>

Generated by Cheméo on 2024-04-25 22:02:23.799336744 +0000 UTC m=+16371792.719914059.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.