

# Androstan-17-one, 3-(formyloxy)-, (3«alpha»,5«beta»)-

**Other names:** 5«beta»-Androstan-17-one, 3«alpha»-hydroxy-, formate

Etiocholan-3«alpha»-ol-17-one, formate

**Inchi:** InChI=1S/C20H30O3/c1-19-9-7-14(23-12-21)11-13(19)3-4-15-16-5-6-18(22)20(16,2)10-8

**InchiKey:** NLSNLCXINBNSRI-AQADWMDFSA-N

**Formula:** C20H30O3

**SMILES:** CC12CCC3C(CCC4CC(OC=O)CCC43C)C1CCC2=O

**Mol. weight [g/mol]:** 318.45

**CAS:** 4589-75-7

## Physical Properties

Property code	Value	Unit	Source
gf	-61.20	kJ/mol	Joback Method
hf	-581.77	kJ/mol	Joback Method
hfus	23.20	kJ/mol	Joback Method
hvap	70.77	kJ/mol	Joback Method
log10ws	-4.58		Crippen Method
logp	4.140		Crippen Method
mcvol	258.230	ml/mol	McGowan Method
pc	1721.73	kPa	Joback Method
rinpol	2324.00		NIST Webbook
rinpol	2324.00		NIST Webbook
tb	830.68	K	Joback Method
tc	1076.08	K	Joback Method
tf	536.85	K	Joback Method
vc	0.979	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	912.89	J/molxK	830.68	Joback Method
cpg	939.50	J/molxK	871.58	Joback Method
cpg	965.71	J/molxK	912.48	Joback Method
cpg	991.87	J/molxK	953.38	Joback Method
cpg	1018.30	J/molxK	994.28	Joback Method

cpg	1045.36	J/mol×K	1035.18	Joback Method
cpg	1073.36	J/mol×K	1076.08	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C4589757&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C4589757&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpolar:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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