

Carbonic acid, but-2-yn-1-yl pentadecyl ester

Inchi: InChI=1S/C20H36O3/c1-3-5-7-8-9-10-11-12-13-14-15-16-17-19-23-20(21)22-18-6-4-2/h3
InchiKey: WBJRILGOHMFYFZ-UHFFFAOYSA-N
Formula: C20H36O3
SMILES: CC#CCOC(=O)OCCCCCCCCCCCCCCC
Mol. weight [g/mol]: 324.50

Physical Properties

Property code	Value	Unit	Source
gf	-18.60	kJ/mol	Joback Method
hf	-560.85	kJ/mol	Joback Method
hfus	54.65	kJ/mol	Joback Method
hvap	73.83	kJ/mol	Joback Method
log10ws	-6.92		Crippen Method
logp	6.254		Crippen Method
mvol	297.370	ml/mol	McGowan Method
pc	1149.87	kPa	Joback Method
rinpol	2328.00		NIST Webbook
rinpol	2328.00		NIST Webbook
tb	764.71	K	Joback Method
tc	946.72	K	Joback Method
tf	515.65	K	Joback Method
vc	1.159	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	895.17	J/mol×K	764.71	Joback Method
cpg	913.91	J/mol×K	795.05	Joback Method
cpg	931.69	J/mol×K	825.38	Joback Method
cpg	948.53	J/mol×K	855.72	Joback Method
cpg	964.42	J/mol×K	886.05	Joback Method
cpg	979.40	J/mol×K	916.39	Joback Method
cpg	993.48	J/mol×K	946.72	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U383210&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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