

# 3-Chloro-2-buten-1-ol

<b>Other names:</b>	2-Buten-1-ol, 3-chloro-3-Chloro-2-butene-1-ol
<b>Inchi:</b>	InChI=1S/C4H7ClO/c1-4(5)2-3-6/h2,6H,3H2,1H3/b4-2-
<b>InchiKey:</b>	SRQGZQPUPABHCN-RQOWECAXSA-N
<b>Formula:</b>	C4H7ClO
<b>SMILES:</b>	CC(Cl)=CCO
<b>Mol. weight [g/mol]:</b>	106.55
<b>CAS:</b>	40605-42-3

## Physical Properties

Property code	Value	Unit	Source
gf	-94.28	kJ/mol	Joback Method
hf	-186.43	kJ/mol	Joback Method
hfus	13.29	kJ/mol	Joback Method
hvap	45.60	kJ/mol	Joback Method
log10ws	-1.26		Crippen Method
logp	1.121		Crippen Method
mvol	81.030	ml/mol	McGowan Method
pc	4504.30	kPa	Joback Method
tb	424.57	K	Joback Method
tc	605.66	K	Joback Method
tf	206.54	K	Joback Method
vc	0.308	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	140.67	J/mol×K	424.57	Joback Method
cpg	147.14	J/mol×K	454.75	Joback Method
cpg	153.26	J/mol×K	484.93	Joback Method
cpg	159.06	J/mol×K	515.12	Joback Method
cpg	164.55	J/mol×K	545.30	Joback Method
cpg	169.75	J/mol×K	575.48	Joback Method
cpg	174.68	J/mol×K	605.66	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	2.27947e+01
Coeff. B	-5.12204e+03
Coeff. C	-1.09720e+02
Temperature range (K), min.	337.29
Temperature range (K), max.	402.69

## Sources

The Yaws Handbook of Vapor Pressure:  
Crippen Method:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

Crippen Method:

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

Joback Method:

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

McGowan Method:

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

NIST Webbook:

<http://link.springer.com/article/10.1007/BF02311772>

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C40605423&Units=SI>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure

**tb:** Normal Boiling Point Temperature  
**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/80-709-6/3-Chloro-2-buten-1-ol.pdf>

Generated by Cheméo on 2024-04-20 05:59:17.972009728 +0000 UTC m=+15882006.892587040.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.