

# 1-Butanol, 2-iodo-

<b>Other names:</b>	2-iodo-1-butanol
<b>Inchi:</b>	InChI=1S/C4H9IO/c1-2-4(5)3-6/h4,6H,2-3H2,1H3
<b>InchiKey:</b>	GVYHZUNZPGWEPZ-UHFFFAOYSA-N
<b>Formula:</b>	C4H9IO
<b>SMILES:</b>	CCC(I)CO
<b>Mol. weight [g/mol]:</b>	200.02
<b>CAS:</b>	127201-28-9

## Physical Properties

Property code	Value	Unit	Source
gf	-98.34	kJ/mol	Joback Method
hf	-206.53	kJ/mol	Joback Method
hfus	11.09	kJ/mol	Joback Method
hvap	50.16	kJ/mol	Joback Method
log10ws	-1.82		Crippen Method
logp	1.192		Crippen Method
mcvol	98.910	ml/mol	McGowan Method
pc	4305.56	kPa	Joback Method
tb	475.80	K	Joback Method
tc	674.21	K	Joback Method
tf	238.72	K	Joback Method
vc	0.360	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	174.03	J/molxK	475.80	Joback Method
cpg	206.84	J/molxK	641.14	Joback Method
cpg	200.99	J/molxK	608.07	Joback Method
cpg	194.79	J/molxK	575.00	Joback Method
cpg	188.25	J/molxK	541.94	Joback Method
cpg	181.33	J/molxK	508.87	Joback Method
cpg	212.39	J/molxK	674.21	Joback Method
dvisc	0.0002665	Paxs	475.80	Joback Method

dvisc	0.0004542	Paxs	436.29	Joback Method
dvisc	0.0008606	Paxs	396.77	Joback Method
dvisc	0.0018783	Paxs	357.26	Joback Method
dvisc	0.0049775	Paxs	317.75	Joback Method
dvisc	0.0173977	Paxs	278.23	Joback Method
dvisc	0.0920207	Paxs	238.72	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C127201289&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C127201289&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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