

# 1,2-Propanediol, 3-methoxy-

<b>Other names:</b>	(.+-.)-3-methoxy-1,2-propanediol 1-O-methyl-rac-glycerol 3-methoxy-1,2-propanediol 3-methoxypropane-1,2-diol DL-1,2-propanediol, 3-methoxy- Glycerin-«alpha»-monomethyl ether Glycerol 1-monomethyl ether NSC 6752
<b>Inchi:</b>	InChI=1S/C4H10O3/c1-7-3-4(6)2-5/h4-6H,2-3H2,1H3
<b>InchiKey:</b>	PSJBSUHYCGQTHZ-UHFFFAOYSA-N
<b>Formula:</b>	C4H10O3
<b>SMILES:</b>	COCC(O)CO
<b>Mol. weight [g/mol]:</b>	106.12
<b>CAS:</b>	623-39-2

## Physical Properties

Property code	Value	Unit	Source
gf	-398.28	kJ/mol	Joback Method
hf	-567.85	kJ/mol	Joback Method
hfus	11.96	kJ/mol	Joback Method
hvap	59.88	kJ/mol	Joback Method
log10ws	0.78		Crippen Method
logp	-1.014		Crippen Method
mcvol	84.830	ml/mol	McGowan Method
pc	4910.80	kPa	Joback Method
rinpola	977.00		NIST Webbook
rinpola	977.00		NIST Webbook
tb	497.26	K	Joback Method
tc	658.88	K	Joback Method
tf	263.71	K	Joback Method
vc	0.309	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	193.62	J/mol×K	497.26	Joback Method
cpg	224.52	J/mol×K	631.94	Joback Method
cpg	218.76	J/mol×K	605.01	Joback Method
cpg	212.79	J/mol×K	578.07	Joback Method
cpg	206.61	J/mol×K	551.13	Joback Method
cpg	200.22	J/mol×K	524.20	Joback Method
cpg	230.06	J/mol×K	658.88	Joback Method
dvisc	0.0000872	Paxs	497.26	Joback Method
dvisc	0.0001812	Paxs	458.33	Joback Method
dvisc	0.0004316	Paxs	419.41	Joback Method
dvisc	0.0012280	Paxs	380.49	Joback Method
dvisc	0.0044333	Paxs	341.56	Joback Method
dvisc	0.0222685	Paxs	302.64	Joback Method
dvisc	0.1801305	Paxs	263.71	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C623392&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C623392&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Solubilities and Thermodynamic Properties of NH3 in Glycerin and its Derivatives:</b>	<a href="https://www.doi.org/10.1021/acs.jced.8b01042">https://www.doi.org/10.1021/acs.jced.8b01042</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices

**tb:** Normal Boiling Point Temperature  
**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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