Oxazole, 2-ethyl-4,5-dihydro-

Other names:	2-Ethyl-2-oxazoline
	2-Ethyloxazoline
	2-Oxazoline, 2-ethyl-
Inchi:	InChI=1S/C5H9NO/c1-2-5-6-3-4-7-5/h2-4H2,1H3
InchiKey:	NYEZZYQZRQDLEH-UHFFFAOYSA-N
Formula:	C5H9NO
SMILES:	CCC1=NCCO1
Mol. weight [g/mol]:	99.13
CAS:	10431-98-8

Physical Properties

Property code	Value	Unit	Source
gf	86.47	kJ/mol	Joback Method
hf	-80.43	kJ/mol	Joback Method
hfus	15.52	kJ/mol	Joback Method
hvap	38.96	kJ/mol	Joback Method
log10ws	-0.56		Crippen Method
logp	0.825		Crippen Method
mcvol	82.000	ml/mol	McGowan Method
pc	4615.13	kPa	Joback Method
rinpol	759.00		NIST Webbook
tb	418.54	K	Joback Method
tc	634.85	K	Joback Method
tf	272.64	K	Joback Method
vc	0.314	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	165.24	J/mol×K	418.54	Joback Method
cpg	177.42	J/mol×K	454.59	Joback Method
cpg	189.00	J/mol×K	490.64	Joback Method
cpg	200.01	J/mol×K	526.69	Joback Method
cpg	210.44	J/mol×K	562.74	Joback Method

cpg	220.30	J/mol×K	598.80	Joback Method
cpg	229.59	J/mol×K	634.85	Joback Method

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C10431988&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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