

# Ethanone, 1-(3-methyloxiranyl)-

<b>Other names:</b>	2-Pentanone, 3,4-epoxy- 3,4-Epoxy-2-pentanone
<b>Inchi:</b>	InChI=1S/C5H8O2/c1-3(6)5-4(2)7-5/h4-5H,1-2H3
<b>InchiKey:</b>	IAHUKWSQLBRAIN-UHFFFAOYSA-N
<b>Formula:</b>	C5H8O2
<b>SMILES:</b>	CC(=O)C1OC1C
<b>Mol. weight [g/mol]:</b>	100.12
<b>CAS:</b>	17257-79-3

## Physical Properties

Property code	Value	Unit	Source
gf	-170.78	kJ/mol	Joback Method
hf	-338.65	kJ/mol	Joback Method
hfus	17.49	kJ/mol	Joback Method
hvap	37.58	kJ/mol	Joback Method
log10ws	-0.40		Crippen Method
logp	0.363		Crippen Method
mvol	77.890	ml/mol	McGowan Method
pc	4178.49	kPa	Joback Method
ripol	1263.00		NIST Webbook
ripol	1263.00		NIST Webbook
tb	396.69	K	Joback Method
tc	591.36	K	Joback Method
tf	236.31	K	Joback Method
vc	0.298	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	152.21	J/molxK	396.69	Joback Method
cpg	162.27	J/molxK	429.14	Joback Method
cpg	171.81	J/molxK	461.58	Joback Method
cpg	180.84	J/molxK	494.03	Joback Method
cpg	189.39	J/molxK	526.47	Joback Method

cpg	197.47	J/mol×K	558.92	Joback Method
cpg	205.12	J/mol×K	591.36	Joback Method
dvisc	0.0011617	Paxs	236.31	Joback Method
dvisc	0.0009407	Paxs	263.04	Joback Method
dvisc	0.0007920	Paxs	289.77	Joback Method
dvisc	0.0006864	Paxs	316.50	Joback Method
dvisc	0.0006083	Paxs	343.23	Joback Method
dvisc	0.0005486	Paxs	369.96	Joback Method
dvisc	0.0005017	Paxs	396.69	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C17257793&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C17257793&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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