

# Propanoic acid, 2-oxo-, ethyl ester

<b>Other names:</b>	Pyruvic acid, ethyl ester Ethyl pyruvate Ethyl 2-oxopropanoate Ethyl 2-oxopropionate
<b>Inchi:</b>	InChI=1S/C5H8O3/c1-3-8-5(7)4(2)6/h3H2,1-2H3
<b>InchiKey:</b>	XXRCUYVCPSWGCC-UHFFFAOYSA-N
<b>Formula:</b>	C5H8O3
<b>SMILES:</b>	CCOC(=O)C(C)=O
<b>Mol. weight [g/mol]:</b>	116.12
<b>CAS:</b>	617-35-6

## Physical Properties

Property code	Value	Unit	Source
gf	-371.62	kJ/mol	Joback Method
hf	-503.91	kJ/mol	Joback Method
hfus	13.09	kJ/mol	Joback Method
hvap	42.63	kJ/mol	Joback Method
log10ws	-0.06		Crippen Method
logp	0.138		Crippen Method
mcvol	90.320	ml/mol	McGowan Method
pc	3960.52	kPa	Joback Method
rinpol	785.00		NIST Webbook
rinpol	785.00		NIST Webbook
rinpol	774.00		NIST Webbook
rinpol	774.00		NIST Webbook
rinpol	774.00		NIST Webbook
ripol	1253.00		NIST Webbook
ripol	1267.00		NIST Webbook
ripol	1242.00		NIST Webbook
ripol	1252.00		NIST Webbook
ripol	1242.00		NIST Webbook
ripol	1242.00		NIST Webbook
ripol	1253.00		NIST Webbook
ripol	1234.00		NIST Webbook
ripol	1276.00		NIST Webbook
ripol	1268.00		NIST Webbook
ripol	1252.00		NIST Webbook

ripol	1267.00		NIST Webbook
ripol	1234.00		NIST Webbook
tb	428.70 ± 0.80	K	NIST Webbook
tb	417.20	K	NIST Webbook
tb	428.20	K	NIST Webbook
tc	634.74	K	Joback Method
tf	268.20	K	Joback Method
vc	0.345	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	176.41	J/mol×K	443.96	Joback Method
cpg	213.74	J/mol×K	602.95	Joback Method
cpg	206.83	J/mol×K	571.15	Joback Method
cpg	199.65	J/mol×K	539.35	Joback Method
cpg	192.17	J/mol×K	507.55	Joback Method
cpg	184.43	J/mol×K	475.76	Joback Method
cpg	220.34	J/mol×K	634.74	Joback Method
dvisc	0.0003266	Paxs	443.96	Joback Method
dvisc	0.0004079	Paxs	414.67	Joback Method
dvisc	0.0005268	Paxs	385.37	Joback Method
dvisc	0.0007096	Paxs	356.08	Joback Method
dvisc	0.0010084	Paxs	326.79	Joback Method
dvisc	0.0015356	Paxs	297.49	Joback Method
dvisc	0.0025634	Paxs	268.20	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C617356&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C617356&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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