

# 4,5-Dihydrofuran-3-carbaldehyde

Inchi:	InChI=1S/C5H6O2/c6-3-5-1-2-7-4-5/h3-4H,1-2H2
InchiKey:	KGQQYFHAJQGVSX-UHFFFAOYSA-N
Formula:	C5H6O2
SMILES:	O=CC1=COCC1
Mol. weight [g/mol]:	98.10

## Physical Properties

Property code	Value	Unit	Source
gf	-129.83	kJ/mol	Joback Method
hf	-236.98	kJ/mol	Joback Method
hfus	12.67	kJ/mol	Joback Method
hvap	39.47	kJ/mol	Joback Method
log10ws	-0.53		Crippen Method
logp	0.489		Crippen Method
mcvol	73.590	ml/mol	McGowan Method
pc	5109.34	kPa	Joback Method
ripol	1172.00		NIST Webbook
ripol	1172.00		NIST Webbook
tb	413.50	K	Joback Method
tc	623.77	K	Joback Method
tf	243.10	K	Joback Method
vc	0.281	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	136.07	J/molxK	413.50	Joback Method
cpg	145.02	J/molxK	448.54	Joback Method
cpg	153.43	J/molxK	483.59	Joback Method
cpg	161.34	J/molxK	518.63	Joback Method
cpg	168.76	J/molxK	553.68	Joback Method
cpg	175.71	J/molxK	588.72	Joback Method
cpg	182.21	J/molxK	623.77	Joback Method
dvisc	0.0034097	Paxs	243.10	Joback Method

dvisc	0.0020238	Paxs	271.50	Joback Method
dvisc	0.0013260	Paxs	299.90	Joback Method
dvisc	0.0009347	Paxs	328.30	Joback Method
dvisc	0.0006966	Paxs	356.70	Joback Method
dvisc	0.0005422	Paxs	385.10	Joback Method
dvisc	0.0004367	Paxs	413.50	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R519090&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R519090&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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