

# Piperazine, 2-methyl-

|                      |                                                   |
|----------------------|---------------------------------------------------|
| Other names:         | 2-Methylpiperazine                                |
| Inchi:               | InChI=1S/C5H12N2/c1-5-4-6-2-3-7-5/h5-7H,2-4H2,1H3 |
| InchiKey:            | JOMNTHCQHJPVAZ-UHFFFAOYSA-N                       |
| Formula:             | C5H12N2                                           |
| SMILES:              | CC1CNCCN1                                         |
| Mol. weight [g/mol]: | 100.16                                            |
| CAS:                 | 109-07-9                                          |

## Physical Properties

| Property code | Value   | Unit    | Source                               |
|---------------|---------|---------|--------------------------------------|
| gf            | 191.09  | kJ/mol  | Joback Method                        |
| hf            | -16.59  | kJ/mol  | Joback Method                        |
| hfus          | 19.72   | kJ/mol  | Joback Method                        |
| hvap          | 40.67   | kJ/mol  | Joback Method                        |
| log10ws       | 0.74    |         | Aqueous Solubility Prediction Method |
| logp          | -0.432  |         | Crippen Method                       |
| mcvol         | 90.410  | ml/mol  | McGowan Method                       |
| pc            | 4802.50 | kPa     | Joback Method                        |
| tb            | 430.45  | K       | Joback Method                        |
| tc            | 655.53  | K       | Joback Method                        |
| tf            | 363.55  | K       | Joback Method                        |
| vc            | 0.323   | m3/kmol | Joback Method                        |

## Temperature Dependent Properties

| Property code | Value  | Unit    | Temperature [K] | Source        |
|---------------|--------|---------|-----------------|---------------|
| cpg           | 174.54 | J/mol×K | 430.45          | Joback Method |
| cpg           | 188.89 | J/mol×K | 467.96          | Joback Method |
| cpg           | 202.62 | J/mol×K | 505.48          | Joback Method |
| cpg           | 215.73 | J/mol×K | 542.99          | Joback Method |
| cpg           | 228.22 | J/mol×K | 580.50          | Joback Method |
| cpg           | 240.09 | J/mol×K | 618.02          | Joback Method |
| cpg           | 251.33 | J/mol×K | 655.53          | Joback Method |

# Pressure Dependent Properties

| Property code | Value  | Unit | Pressure [kPa] | Source       |
|---------------|--------|------|----------------|--------------|
| tbrp          | 428.20 | K    | 102.00         | NIST Webbook |

## Correlations

| Information                 | Value                         |
|-----------------------------|-------------------------------|
| Property code               | pvap                          |
| Equation                    | $\ln(P_{vp}) = A + B/(T + C)$ |
| Coeff. A                    | 1.53054e+01                   |
| Coeff. B                    | -3.93903e+03                  |
| Coeff. C                    | -5.96200e+01                  |
| Temperature range (K), min. | 321.91                        |
| Temperature range (K), max. | 453.76                        |

## Sources

|                                       |                                                                                                                                                                                                                         |
|---------------------------------------|-------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|
| Aqueous Solubility Prediction Method: | <a href="http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDataset002.xlsx">http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDataset002.xlsx</a> |
| McGowan Method:                       | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                                                                                                   |
| NIST Webbook:                         | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C109079&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C109079&amp;Units=SI</a>                                                                               |
| The Yaws Handbook of Vapor Pressure:  | <a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>                                 |
| Crippen Method:                       | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>                                                                                                               |
| Joback Method:                        | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                                                                                                   |

## Legend

|       |                                              |
|-------|----------------------------------------------|
| cpg:  | Ideal gas heat capacity                      |
| gf:   | Standard Gibbs free energy of formation      |
| hf:   | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions    |

|                                       |                                                 |
|---------------------------------------|-------------------------------------------------|
| <b>h<sub>vap</sub>:</b>               | Enthalpy of vaporization at standard conditions |
| <b>log<sub>10</sub>w<sub>s</sub>:</b> | Log10 of Water solubility in mol/l              |
| <b>log<sub>p</sub>:</b>               | Octanol/Water partition coefficient             |
| <b>mc<sub>vol</sub>:</b>              | McGowan's characteristic volume                 |
| <b>p<sub>c</sub>:</b>                 | Critical Pressure                               |
| <b>p<sub>vap</sub>:</b>               | Vapor pressure                                  |
| <b>t<sub>b</sub>:</b>                 | Normal Boiling Point Temperature                |
| <b>t<sub>brp</sub>:</b>               | Boiling point at reduced pressure               |
| <b>t<sub>c</sub>:</b>                 | Critical Temperature                            |
| <b>t<sub>f</sub>:</b>                 | Normal melting (fusion) point                   |
| <b>v<sub>c</sub>:</b>                 | Critical Volume                                 |

Latest version available from:

<https://www.cheméo.com/cid/80-851-8/Piperazine-2-methyl.pdf>

Generated by Cheméo on 2025-12-05 14:42:26.714779283 +0000 UTC m=+4693944.244819947.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.