

# Ethane, 1-ethoxy-1-methoxy-

<b>Other names:</b>	Acetaldehyde, ethyl methyl acetal Acetaldehyde methyl ethyl acetyl 1,1-Ethoxymethoxyethane 1-ethoxy-1-methoxyethane
<b>Inchi:</b>	InChI=1S/C5H12O2/c1-4-7-5(2)6-3/h5H,4H2,1-3H3
<b>InchiKey:</b>	MYCXIDKAJBXPCZ-UHFFFAOYSA-N
<b>Formula:</b>	C5H12O2
<b>SMILES:</b>	CCOC(C)OC
<b>Mol. weight [g/mol]:</b>	104.15
<b>CAS:</b>	10471-14-4

## Physical Properties

Property code	Value	Unit	Source
gf	-221.22	kJ/mol	Joback Method
hf	-416.25	kJ/mol	Joback Method
hfus	7.56	kJ/mol	Joback Method
hvap	31.16	kJ/mol	Joback Method
log10ws	-0.70		Crippen Method
logp	1.015		Crippen Method
mcvol	93.050	ml/mol	McGowan Method
pc	3364.54	kPa	Joback Method
rinpol	641.00		NIST Webbook
rinpol	641.00		NIST Webbook
tb	358.20	K	Joback Method
tc	527.59	K	Joback Method
tf	175.57	K	Joback Method
vc	0.345	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	169.41	J/mol×K	358.20	Joback Method
cpg	178.19	J/mol×K	386.43	Joback Method
cpg	186.80	J/mol×K	414.66	Joback Method

cpg	195.24	J/molxK	442.89	Joback Method
cpg	203.50	J/molxK	471.12	Joback Method
cpg	211.58	J/molxK	499.35	Joback Method
cpg	219.47	J/molxK	527.59	Joback Method
dvisc	0.0043341	Paxs	175.57	Joback Method
dvisc	0.0017713	Paxs	206.01	Joback Method
dvisc	0.0009115	Paxs	236.45	Joback Method
dvisc	0.0005458	Paxs	266.88	Joback Method
dvisc	0.0003630	Paxs	297.32	Joback Method
dvisc	0.0002604	Paxs	327.76	Joback Method
dvisc	0.0001977	Paxs	358.20	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C10471144&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C10471144&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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