

Ethane, 1,1,2-trimethoxy-

Other names:	Methoxyacetaldehyde dimethyl acetal 2-Methoxyacetaldehyde dimethyl acetal 1,1,2-Trimethoxyethane
Inchi:	InChI=1S/C5H12O3/c1-6-4-5(7-2)8-3/h5H,4H2,1-3H3
InchiKey:	DYOZNCVZPFIKLU-UHFFFAOYSA-N
Formula:	C5H12O3
SMILES:	COCC(OC)OC
Mol. weight [g/mol]:	120.15
CAS:	24332-20-5

Physical Properties

Property code	Value	Unit	Source
gf	-326.22	kJ/mol	Joback Method
hf	-548.47	kJ/mol	Joback Method
hfus	8.75	kJ/mol	Joback Method
hvap	33.57	kJ/mol	Joback Method
log10ws	0.22		Crippen Method
logp	0.252		Crippen Method
mcvol	98.920	ml/mol	McGowan Method
pc	3299.15	kPa	Joback Method
tb	380.62	K	Joback Method
tc	550.72	K	Joback Method
tf	197.80	K	Joback Method
vc	0.363	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	190.90	J/mol×K	380.62	Joback Method
cpg	199.54	J/mol×K	408.97	Joback Method
cpg	208.09	J/mol×K	437.32	Joback Method
cpg	216.53	J/mol×K	465.67	Joback Method
cpg	224.83	J/mol×K	494.02	Joback Method
cpg	232.99	J/mol×K	522.37	Joback Method

cpg	240.98	J/molxK	550.72	Joback Method
dvisc	0.0031901	Paxs	197.80	Joback Method
dvisc	0.0014335	Paxs	228.27	Joback Method
dvisc	0.0007777	Paxs	258.74	Joback Method
dvisc	0.0004800	Paxs	289.21	Joback Method
dvisc	0.0003248	Paxs	319.68	Joback Method
dvisc	0.0002352	Paxs	350.15	Joback Method
dvisc	0.0001794	Paxs	380.62	Joback Method

Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	330.70	K	7.50	NIST Webbook

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C24332205&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature

tbrp: Boiling point at reduced pressure
tc: Critical Temperature
tf: Normal melting (fusion) point
vc: Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/80-865-3/Ethane-1-1-2-trimethoxy.pdf>

Generated by Cheméo on 2024-04-18 22:21:00.996551206 +0000 UTC m=+15768109.917128541.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.