

# ethyl 4-hydroxybutanoate

<b>Other names:</b>	ethyl 4-hydroxybutyrate
<b>Inchi:</b>	InChI=1S/C6H12O3/c1-2-9-6(8)4-3-5-7/h7H,2-5H2,1H3
<b>InchiKey:</b>	AYPJVXQBVHCUCJ-UHFFFAOYSA-N
<b>Formula:</b>	C6H12O3
<b>SMILES:</b>	CCOC(=O)CCCO
<b>Mol. weight [g/mol]:</b>	132.16

## Physical Properties

Property code	Value	Unit	Source
gf	-371.10	kJ/mol	Joback Method
hf	-564.20	kJ/mol	Joback Method
hfus	18.17	kJ/mol	Joback Method
hvap	54.78	kJ/mol	Joback Method
log10ws	-0.46		Crippen Method
logp	0.322		Crippen Method
mcvol	108.710	ml/mol	McGowan Method
pc	3646.53	kPa	Joback Method
rinpol	1022.00		NIST Webbook
rinpol	1022.00		NIST Webbook
rinpol	1022.00		NIST Webbook
rinpol	1022.00		NIST Webbook
rinpol	1039.00		NIST Webbook
ripol	1819.00		NIST Webbook
ripol	1819.00		NIST Webbook
ripol	1794.00		NIST Webbook
ripol	1819.00		NIST Webbook
ripol	1819.00		NIST Webbook
ripol	1790.00		NIST Webbook
ripol	1822.00		NIST Webbook
ripol	1794.00		NIST Webbook
ripol	1796.00		NIST Webbook
ripol	1795.00		NIST Webbook
ripol	1790.00		NIST Webbook
ripol	1819.00		NIST Webbook
tb	505.15	K	Joback Method
tc	676.09	K	Joback Method
tf	290.36	K	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	245.47	J/mol×K	505.15	Joback Method
cpg	254.32	J/mol×K	533.64	Joback Method
cpg	262.85	J/mol×K	562.13	Joback Method
cpg	271.06	J/mol×K	590.62	Joback Method
cpg	278.97	J/mol×K	619.11	Joback Method
cpg	286.56	J/mol×K	647.60	Joback Method
cpg	293.84	J/mol×K	676.09	Joback Method
dvisc	0.0120283	Paxs	290.36	Joback Method
dvisc	0.0039241	Paxs	326.16	Joback Method
dvisc	0.0015977	Paxs	361.96	Joback Method
dvisc	0.0007647	Paxs	397.75	Joback Method
dvisc	0.0004134	Paxs	433.55	Joback Method
dvisc	0.0002454	Paxs	469.35	Joback Method
dvisc	0.0001569	Paxs	505.15	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R182108&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R182108&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions

<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>p<sub>c</sub>:</b>	Critical Pressure
<b>r<sub>inpol</sub>:</b>	Non-polar retention indices
<b>r<sub>ipol</sub>:</b>	Polar retention indices
<b>t<sub>b</sub>:</b>	Normal Boiling Point Temperature
<b>t<sub>c</sub>:</b>	Critical Temperature
<b>t<sub>f</sub>:</b>	Normal melting (fusion) point
<b>v<sub>c</sub>:</b>	Critical Volume

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