

Acetic acid, ethoxy-, ethyl ester

Other names:	Ethoxyacetic acid ethyl ester Ethyl ethoxyacetate Ethyl 2-ethoxyacetate EThyl ester of ethoxyacetic acid
Inchi:	InChI=1S/C6H12O3/c1-3-8-5-6(7)9-4-2/h3-5H2,1-2H3
InchiKey:	CKSRFHWWBKRUKA-UHFFFAOYSA-N
Formula:	C6H12O3
SMILES:	CCOCC(=O)OCC
Mol. weight [g/mol]:	132.16
CAS:	817-95-8

Physical Properties

Property code	Value	Unit	Source
chl	-3436.50 ± 0.60	kJ/mol	NIST Webbook
gf	-339.28	kJ/mol	Joback Method
hf	-544.19	kJ/mol	Joback Method
hfl	-639.50 ± 0.60	kJ/mol	NIST Webbook
hfus	15.27	kJ/mol	Joback Method
hvap	40.52	kJ/mol	Joback Method
log10ws	-0.28		Crippen Method
logp	0.586		Crippen Method
mcvol	108.710	ml/mol	McGowan Method
pc	3210.04	kPa	Joback Method
tb	435.39	K	Joback Method
tc	612.77	K	Joback Method
tf	251.77	K	Joback Method
vc	0.413	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	276.12	J/mol×K	612.77	Joback Method
cpg	267.77	J/mol×K	583.20	Joback Method
cpg	259.12	J/mol×K	553.64	Joback Method

cpg	250.20	J/molxK	524.08	Joback Method
cpg	241.01	J/molxK	494.52	Joback Method
cpg	231.56	J/molxK	464.95	Joback Method
cpg	221.86	J/molxK	435.39	Joback Method
dvisc	0.0023571	Paxs	251.77	Joback Method
dvisc	0.0002392	Paxs	435.39	Joback Method
dvisc	0.0003033	Paxs	404.79	Joback Method
dvisc	0.0003996	Paxs	374.18	Joback Method
dvisc	0.0005532	Paxs	343.58	Joback Method
dvisc	0.0008159	Paxs	312.98	Joback Method
dvisc	0.0013093	Paxs	282.37	Joback Method
hvapt	46.10	kJ/mol	380.00	NIST Webbook

Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	333.50 ± 0.50	K	2.40	NIST Webbook

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C817958&Units=SI

Legend

chl:	Standard liquid enthalpy of combustion
cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfl:	Liquid phase enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions

h_{vap}:	Enthalpy of vaporization at standard conditions
h_{vapt}:	Enthalpy of vaporization at a given temperature
log₁₀w_s:	Log10 of Water solubility in mol/l
log_p:	Octanol/Water partition coefficient
m_{cvol}:	McGowan's characteristic volume
p_c:	Critical Pressure
t_b:	Normal Boiling Point Temperature
t_{brp}:	Boiling point at reduced pressure
t_c:	Critical Temperature
t_f:	Normal melting (fusion) point
v_c:	Critical Volume

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