

# Pentane, 1-bromo-2-methoxy

<b>Inchi:</b>	InChI=1S/C6H13BrO/c1-3-4-6(5-7)8-2/h6H,3-5H2,1-2H3
<b>InchiKey:</b>	JDAILCRIKVLCFU-UHFFFAOYSA-N
<b>Formula:</b>	C6H13BrO
<b>SMILES:</b>	CCCC(CBr)OC
<b>Mol. weight [g/mol]:</b>	181.07

## Physical Properties

Property code	Value	Unit	Source
gf	-93.48	kJ/mol	Joback Method
hf	-278.34	kJ/mol	Joback Method
hfus	14.25	kJ/mol	Joback Method
hvap	37.41	kJ/mol	Joback Method
log10ws	-1.96		Crippen Method
logp	2.196		Crippen Method
mcvol	118.770	ml/mol	McGowan Method
pc	3325.84	kPa	Joback Method
rinpol	936.00		NIST Webbook
rinpol	936.00		NIST Webbook
tb	424.82	K	Joback Method
tc	612.54	K	Joback Method
tf	224.41	K	Joback Method
vc	0.446	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	220.06	J/molxK	424.82	Joback Method
cpg	269.61	J/molxK	581.26	Joback Method
cpg	260.50	J/molxK	549.97	Joback Method
cpg	251.00	J/molxK	518.68	Joback Method
cpg	241.10	J/molxK	487.39	Joback Method
cpg	230.79	J/molxK	456.11	Joback Method
cpg	278.33	J/molxK	612.54	Joback Method
dvisc	0.0002837	Paxs	424.82	Joback Method

dvisc	0.0003698	Paxs	391.42	Joback Method
dvisc	0.0005066	Paxs	358.02	Joback Method
dvisc	0.0007403	Paxs	324.62	Joback Method
dvisc	0.0011802	Paxs	291.21	Joback Method
dvisc	0.0021234	Paxs	257.81	Joback Method
dvisc	0.0045501	Paxs	224.41	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R12024&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R12024&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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