

3-METHOXY-3-METHYL-1-BUTANOL

Other names:	1-Butanol, 3-methoxy-3-methyl- 3-Methyl-3-methoxy-1-butanol 3-methoxy-3-methylbutan-1-ol 3-methoxy-3-methylbutanol
Inchi:	InChI=1S/C6H14O2/c1-6(2,8-3)4-5-7/h7H,4-5H2,1-3H3
InchiKey:	MFKRHJVUCZRDTF-UHFFFAOYSA-N
Formula:	C6H14O2
SMILES:	COC(C)(C)CCO
Mol. weight [g/mol]:	118.18
CAS:	56539-66-3

Physical Properties

Property code	Value	Unit	Source
hf	-473.56	kJ/mol	Cheméo Relay (1.0)
hfus	9.16	kJ/mol	Joback Method
hvap	58.48	kJ/mol	Cheméo Relay (1.0)
log10ws	0.64		Cheméo Relay (1.0)
logp	0.794		Crippen Method
mcvol	107.140	ml/mol	McGowan Method
ripol	1434.00		NIST Webbook
ripol	1434.00		NIST Webbook
tb	443.69	K	Cheméo Relay (1.0)

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	232.93	J/mol×K	448.05	Joback Method
cpg	243.25	J/mol×K	476.53	Joback Method
cpg	253.13	J/mol×K	505.01	Joback Method
cpg	262.58	J/mol×K	533.49	Joback Method
cpg	271.62	J/mol×K	561.97	Joback Method
cpg	280.27	J/mol×K	590.46	Joback Method
cpg	288.52	J/mol×K	618.94	Joback Method
dvisc	0.0527584	Pa×s	242.85	Joback Method

dvisc	0.0117321	Pa×s	277.05	Joback Method
dvisc	0.0036303	Pa×s	311.25	Joback Method
dvisc	0.0014170	Pa×s	345.45	Joback Method
dvisc	0.0006553	Pa×s	379.65	Joback Method
dvisc	0.0003442	Pa×s	413.85	Joback Method
dvisc	0.0001995	Pa×s	448.05	Joback Method

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.87890e+01
Coeff. B	-5.15009e+03
Coeff. C	-6.54580e+01
Temperature range (K), min.	343.82
Temperature range (K), max.	447.58

Sources

Cheméo Relay (1.0):

Cheméo Relay (1.0, beta):

The Yaws Handbook of Vapor
Pressure:
NIST Webbook:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C56539663&Units=SI>

Joback Method:

https://en.wikipedia.org/wiki/Joback_method

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

Crippen Method:

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

Crippen Method:

https://www.chemeo.com/doc/models/crippen_log10ws

Legend

cpg: Ideal gas heat capacity

dvisc: Dynamic viscosity

hf: Enthalpy of formation at standard conditions

hfus: Enthalpy of fusion at standard conditions

h_{vap}:	Enthalpy of vaporization at standard conditions
log₁₀ws:	Log10 of Water solubility in mol/l
log_p:	Octanol/Water partition coefficient
mc_{vol}:	McGowan's characteristic volume
p_{vap}:	Vapor pressure
ri_{pol}:	Polar retention indices
tb:	Normal Boiling Point Temperature

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