

# Diazene, ethyl-(2-methylpropyl)

<b>Other names:</b>	trans-ethyl-i-butyl-diazene
<b>Inchi:</b>	InChI=1S/C6H14N2/c1-4-7-8-5-6(2)3/h6H,4-5H2,1-3H3
<b>InchiKey:</b>	XTJYPGLPPGVKTM-UHFFFAOYSA-N
<b>Formula:</b>	C6H14N2
<b>SMILES:</b>	CCN=NCC(C)C
<b>Mol. weight [g/mol]:</b>	114.19

## Physical Properties

Property code	Value	Unit	Source
hf	-125.23	kJ/mol	Joback Method
hvap	35.23	kJ/mol	Joback Method
log10ws	-1.35		Crippen Method
logp	2.115		Crippen Method
mcvol	111.060	ml/mol	McGowan Method
pc	2384.19	kPa	Joback Method
rinpol	725.00		NIST Webbook
rinpol	731.70		NIST Webbook
tb	485.44	K	Joback Method
tc	688.98	K	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R129627&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R129627&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

hf: Enthalpy of formation at standard conditions

<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log10 of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>p<sub>c</sub>:</b>	Critical Pressure
<b>r<sub>inpol</sub>:</b>	Non-polar retention indices
<b>t<sub>b</sub>:</b>	Normal Boiling Point Temperature
<b>t<sub>c</sub>:</b>	Critical Temperature

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