

2(5H)-Furanone, 3,5,5-trimethyl-

Other names:	2,4,4-Trimethyl-2-butenolide 3,5,5-trimethyl-2(5H)-furanone
Inchi:	InChI=1S/C7H10O2/c1-5-4-7(2,3)9-6(5)8/h4H,1-3H3
InchiKey:	OAVLYQLCBORARJ-UHFFFAOYSA-N
Formula:	C7H10O2
SMILES:	CC1=CC(C)(C)OC1=O
Mol. weight [g/mol]:	126.15
CAS:	50598-50-0

Physical Properties

Property code	Value	Unit	Source
gf	-149.26	kJ/mol	Joback Method
hf	-335.48	kJ/mol	Joback Method
hfus	9.84	kJ/mol	Joback Method
hvap	39.99	kJ/mol	Joback Method
log10ws	-1.48		Crippen Method
logp	1.268		Crippen Method
mcvol	101.770	ml/mol	McGowan Method
pc	3819.82	kPa	Joback Method
ripol	2074.00		NIST Webbook
ripol	2074.00		NIST Webbook
ripol	2074.00		NIST Webbook
tb	473.99	K	Joback Method
tc	701.21	K	Joback Method
tf	311.52	K	Joback Method
vc	0.381	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	219.58	J/molxK	473.99	Joback Method
cpg	232.11	J/molxK	511.86	Joback Method
cpg	243.87	J/molxK	549.73	Joback Method
cpg	254.94	J/molxK	587.60	Joback Method

cpg	265.41	J/mol×K	625.47	Joback Method
cpg	275.38	J/mol×K	663.34	Joback Method
cpg	284.92	J/mol×K	701.21	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C50598500&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mvol:	McGowan's characteristic volume
pc:	Critical Pressure
ripol:	Polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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