

# 2H-Pyran-2-one, tetrahydro-5,6-dimethyl-, trans-

<b>Other names:</b>	trans-4-Methyl-5-Hydroxyhexanoic acid lactone 4-Methyl-5-hexanolide
<b>Inchi:</b>	InChI=1S/C7H12O2/c1-5-3-4-7(8)9-6(5)2/h5-6H,3-4H2,1-2H3/t5-,6+/m0/s1
<b>InchiKey:</b>	HAXARIVGMMVELD-NTSWFWBYSAN
<b>Formula:</b>	C7H12O2
<b>SMILES:</b>	CC1CCC(=O)OC1C
<b>Mol. weight [g/mol]:</b>	128.17
<b>CAS:</b>	24405-16-1

## Physical Properties

Property code	Value	Unit	Source
gf	-183.91	kJ/mol	Joback Method
hf	-423.53	kJ/mol	Joback Method
hfus	14.28	kJ/mol	Joback Method
hvap	40.05	kJ/mol	Joback Method
log10ws	-1.38		Crippen Method
logp	1.348		Crippen Method
mvol	106.070	ml/mol	McGowan Method
pc	3493.01	kPa	Joback Method
rinpol	1133.00		NIST Webbook
rinpol	1133.00		NIST Webbook
tb	469.21	K	Joback Method
tc	691.39	K	Joback Method
tf	266.58	K	Joback Method
vc	0.388	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	234.42	J/molxK	469.21	Joback Method
cpg	250.08	J/molxK	506.24	Joback Method
cpg	265.11	J/molxK	543.27	Joback Method
cpg	279.49	J/molxK	580.30	Joback Method
cpg	293.21	J/molxK	617.33	Joback Method

cpg	306.23	J/mol×K	654.36	Joback Method
cpg	318.55	J/mol×K	691.39	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C24405161&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C24405161&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m cvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>r inpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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