

# Propane, 2,2-bis(tert-butyloxy)-

<b>Inchi:</b>	InChI=1S/C11H24O2/c1-9(2,3)12-11(7,8)13-10(4,5)6/h1-8H3
<b>InchiKey:</b>	MJMULMSFDFSXNW-UHFFFAOYSA-N
<b>Formula:</b>	C11H24O2
<b>SMILES:</b>	CC(C)(C)OC(C)(C)OC(C)(C)C
<b>Mol. weight [g/mol]:</b>	188.31

## Physical Properties

Property code	Value	Unit	Source
gf	-159.74	kJ/mol	Joback Method
hf	-561.06	kJ/mol	Joback Method
hfus	4.38	kJ/mol	Joback Method
hvap	41.01	kJ/mol	Joback Method
log10ws	-3.43		Crippen Method
logp	3.353		Crippen Method
mvol	177.590	ml/mol	McGowan Method
pc	1973.55	kPa	Joback Method
rinpol	1107.00		NIST Webbook
rinpol	1107.00		NIST Webbook
tb	486.23	K	Joback Method
tc	678.19	K	Joback Method
tf	265.45	K	Joback Method
vc	0.654	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	423.29	J/molxK	486.23	Joback Method
cpg	442.46	J/molxK	518.22	Joback Method
cpg	460.57	J/molxK	550.22	Joback Method
cpg	477.66	J/molxK	582.21	Joback Method
cpg	493.78	J/molxK	614.20	Joback Method
cpg	508.98	J/molxK	646.20	Joback Method
cpg	523.29	J/molxK	678.19	Joback Method
dvisc	0.0077725	Paxs	265.45	Joback Method

dvisc	0.0026919	Paxs	302.25	Joback Method
dvisc	0.0011736	Paxs	339.04	Joback Method
dvisc	0.0006019	Paxs	375.84	Joback Method
dvisc	0.0003478	Paxs	412.64	Joback Method
dvisc	0.0002198	Paxs	449.43	Joback Method
dvisc	0.0001489	Paxs	486.23	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U381761&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U381761&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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