

1-Propene, 1,1,3-trichloro-

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|-----------------------------|---|
| Other names: | 1,1,3-Trichloro-1-propene 1,1,3-Trichloropropene 1,3,3-Trichloro-2-propene 3,3-Dichloroallyl chloride Propene, 1,1,3-trichloro- |
| Inchi: | InChI=1S/C3H3Cl3/c4-2-1-3(5)6/h1H,2H2 |
| InchiKey: | JFEVIPGMXQNRRF-UHFFFAOYSA-N |
| Formula: | C3H3Cl3 |
| SMILES: | C1CC=C(Cl)Cl |
| Mol. weight [g/mol]: | 145.41 |
| CAS: | 2567-14-8 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|----------------------|----------------|
| gf | 10.26 | kJ/mol | Joback Method |
| hf | -45.04 | kJ/mol | Joback Method |
| hfus | 15.01 | kJ/mol | Joback Method |
| hvap | 35.47 | kJ/mol | Joback Method |
| log10ws | -2.38 | | Crippen Method |
| logp | 2.544 | | Crippen Method |
| mvol | 85.550 | ml/mol | McGowan Method |
| pc | 4109.14 | kPa | Joback Method |
| rinpol | 840.00 | | NIST Webbook |
| rinpol | 840.00 | | NIST Webbook |
| tb | 404.70 | K | NIST Webbook |
| tc | 590.66 | K | Joback Method |
| tf | 194.29 | K | Joback Method |
| vc | 0.332 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|--------|---------|-----------------|---------------|
| cpg | 114.82 | J/molxK | 384.37 | Joback Method |
| cpg | 119.89 | J/molxK | 418.75 | Joback Method |

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|-----|--------|---------|--------|---------------|
| cpg | 124.59 | J/mol×K | 453.13 | Joback Method |
| cpg | 128.93 | J/mol×K | 487.51 | Joback Method |
| cpg | 132.94 | J/mol×K | 521.89 | Joback Method |
| cpg | 136.65 | J/mol×K | 556.28 | Joback Method |
| cpg | 140.07 | J/mol×K | 590.66 | Joback Method |

Correlations

| Information | Value |
|-----------------------------|-------------------------------|
| Property code | pvap |
| Equation | $\ln(P_{vp}) = A + B/(T + C)$ |
| Coeff. A | 1.42428e+01 |
| Coeff. B | -3.37859e+03 |
| Coeff. C | -5.36570e+01 |
| Temperature range (K), min. | 295.76 |
| Temperature range (K), max. | 431.94 |

Sources

| | |
|---|---|
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=C2567148&Units=SI |
| The Yaws Handbook of Vapor Pressure: | https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci990307l |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |

| | |
|----------------|----------------------------------|
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| pvap: | Vapor pressure |
| rinpol: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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