

# 1,2-Ethanediamine, N,N-bis(1-methylethyl)-

<b>Other names:</b>	Ethylenediamine, N,N-diisopropyl- N,N-Diisopropyl-1,2-ethanediamine N,N-Diisopropylethylenediamine 2-(Diisopropylamino)ethylamine USAF am-2 N,N-bis(1-methylethyl)-1,2-ethanediamine 2-aminoethyldiisopropylamine
<b>Inchi:</b>	InChI=1S/C8H20N2/c1-7(2)10(6-5-9)8(3)4/h7-8H,5-6,9H2,1-4H3
<b>InchiKey:</b>	CURJNMSGPBXOGK-UHFFFAOYSA-N
<b>Formula:</b>	C8H20N2
<b>SMILES:</b>	CC(C)N(CCN)C(C)C
<b>Mol. weight [g/mol]:</b>	144.26
<b>CAS:</b>	121-05-1

## Physical Properties

Property code	Value	Unit	Source
gf	188.83	kJ/mol	Joback Method
hf	-117.69	kJ/mol	Joback Method
hfus	17.65	kJ/mol	Joback Method
hvap	45.31	kJ/mol	Joback Method
log10ws	-1.40		Crippen Method
logp	1.064		Crippen Method
mcvol	143.540	ml/mol	McGowan Method
pc	2726.86	kPa	Joback Method
ripol	1776.00		NIST Webbook
ripol	1776.00		NIST Webbook
tb	466.53	K	Joback Method
tc	649.74	K	Joback Method
tf	265.65	K	Joback Method
vc	0.518	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	325.07	J/mol×K	466.53	Joback Method
cpg	340.34	J/mol×K	497.07	Joback Method
cpg	354.90	J/mol×K	527.60	Joback Method
cpg	368.79	J/mol×K	558.14	Joback Method
cpg	382.01	J/mol×K	588.67	Joback Method
cpg	394.60	J/mol×K	619.21	Joback Method
cpg	406.57	J/mol×K	649.74	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C121051&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C121051&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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