

# Benzoic acid, 3-formyl-

<b>Other names:</b>	3-Carboxybenzaldehyde 3-formylbenzoic acid
<b>Inchi:</b>	InChI=1S/C8H6O3/c9-5-6-2-1-3-7(4-6)8(10)11/h1-5H,(H,10,11)
<b>InchiKey:</b>	UHDNUPHSDMOGCR-UHFFFAOYSA-N
<b>Formula:</b>	C8H6O3
<b>SMILES:</b>	O=Cc1cccc(C(=O)O)c1
<b>Mol. weight [g/mol]:</b>	150.13
<b>CAS:</b>	619-21-6

## Physical Properties

Property code	Value	Unit	Source
gf	-246.00	kJ/mol	Joback Method
hf	-333.78	kJ/mol	Joback Method
hfus	18.10	kJ/mol	Joback Method
hvap	66.48	kJ/mol	Joback Method
log10ws	-1.77		Crippen Method
logp	1.197		Crippen Method
mvol	108.830	ml/mol	McGowan Method
pc	4815.84	kPa	Joback Method
tb	608.81	K	Joback Method
tc	819.05	K	Joback Method
tf	371.61	K	Joback Method
vc	0.417	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	246.78	J/mol×K	608.81	Joback Method
cpg	254.76	J/mol×K	643.85	Joback Method
cpg	262.19	J/mol×K	678.89	Joback Method
cpg	269.11	J/mol×K	713.93	Joback Method
cpg	275.53	J/mol×K	748.97	Joback Method
cpg	281.48	J/mol×K	784.01	Joback Method
cpg	286.98	J/mol×K	819.05	Joback Method

dvisc	0.0034418	Paxs	371.61	Joback Method
dvisc	0.0015021	Paxs	411.14	Joback Method
dvisc	0.0007582	Paxs	450.68	Joback Method
dvisc	0.0004273	Paxs	490.21	Joback Method
dvisc	0.0002624	Paxs	529.74	Joback Method
dvisc	0.0001724	Paxs	569.28	Joback Method
dvisc	0.0001196	Paxs	608.81	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C619216&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C619216&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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