ethyl 2-methyl-2(Z)-pentenoate

Inchi: InChI=1S/C8H14O2/c1-4-6-7(3)8(9)10-5-2/h6H,4-5H2,1-3H3/b7-6-
InchiKey: HYQYCHQAUPHFKX-SREVYHEPSA-N
Formula: C8H14O2
SMILES: CCC=C(C)C(=O)OCC
Mol. weight [g/mol]: 142.20

Physical Properties

Property code	Value	Unit	Source
gf	-145.77	kJ/mol	Joback Method
hf	-345.82	kJ/mol	Joback Method
hfus	18.16	kJ/mol	Joback Method
hvap	42.60	kJ/mol	Joback Method
log10ws	-1.89		Crippen Method
logp	1.906		Crippen Method
mcvol	126.720	ml/mol	McGowan Method
pc	2820.33	kPa	Joback Method
ripol	1149.00		NIST Webbook
ripol	1152.00		NIST Webbook
ripol	1149.00		NIST Webbook
tb	462.77	K	Joback Method
tc	649.52	K	Joback Method
tf	233.04	K	Joback Method
vc	0.488	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	263.21	J/mol×K	462.77	Joback Method
cpg	275.29	J/mol×K	493.90	Joback Method
cpg	286.87	J/mol×K	525.02	Joback Method
cpg	297.94	J/mol×K	556.15	Joback Method
cpg	308.52	J/mol×K	587.27	Joback Method
cpg	318.63	J/mol×K	618.40	Joback Method
cpg	328.27	J/mol×K	649.52	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R315081&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
ripol:	Polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume