

# 2',4'-Dihydroxy-3'-methylacetophenone

|                      |                                                                  |
|----------------------|------------------------------------------------------------------|
| Other names:         | 1-(2,4-Dihydroxy-3-methylphenyl)ethanone                         |
| Inchi:               | InChI=1S/C9H10O3/c1-5-8(11)4-3-7(6(2)10)9(5)12/h3-4,11-12H,1-2H3 |
| InchiKey:            | KMTLZBUHQPPQFAV-UHFFFAOYSA-N                                     |
| Formula:             | C9H10O3                                                          |
| SMILES:              | CC(=O)c1ccc(O)c(C)c1O                                            |
| Mol. weight [g/mol]: | 166.17                                                           |
| CAS:                 | 10139-84-1                                                       |

## Physical Properties

| Property code | Value   | Unit    | Source         |
|---------------|---------|---------|----------------|
| gf            | -310.48 | kJ/mol  | Joback Method  |
| hf            | -471.23 | kJ/mol  | Joback Method  |
| hfus          | 25.88   | kJ/mol  | Joback Method  |
| hvap          | 71.34   | kJ/mol  | Joback Method  |
| log10ws       | -1.70   |         | Crippen Method |
| logp          | 1.609   |         | Crippen Method |
| mcvol         | 127.220 | ml/mol  | McGowan Method |
| pc            | 5065.79 | kPa     | Joback Method  |
| ripol         | 1893.00 |         | NIST Webbook   |
| ripol         | 1893.00 |         | NIST Webbook   |
| tb            | 652.09  | K       | Joback Method  |
| tc            | 893.33  | K       | Joback Method  |
| tf            | 503.50  | K       | Joback Method  |
| vc            | 0.369   | m3/kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value  | Unit    | Temperature [K] | Source        |
|---------------|--------|---------|-----------------|---------------|
| cpg           | 322.95 | J/mol×K | 652.09          | Joback Method |
| cpg           | 332.56 | J/mol×K | 692.30          | Joback Method |
| cpg           | 341.55 | J/mol×K | 732.50          | Joback Method |
| cpg           | 350.07 | J/mol×K | 772.71          | Joback Method |
| cpg           | 358.26 | J/mol×K | 812.91          | Joback Method |
| cpg           | 366.25 | J/mol×K | 853.12          | Joback Method |

|       |           |         |        |               |
|-------|-----------|---------|--------|---------------|
| cpg   | 374.17    | J/mol×K | 893.33 | Joback Method |
| dvisc | 0.0000729 | Pa×s    | 503.50 | Joback Method |
| dvisc | 0.0000401 | Pa×s    | 528.26 | Joback Method |
| dvisc | 0.0000233 | Pa×s    | 553.03 | Joback Method |
| dvisc | 0.0000142 | Pa×s    | 577.80 | Joback Method |
| dvisc | 0.0000090 | Pa×s    | 602.56 | Joback Method |
| dvisc | 0.0000059 | Pa×s    | 627.33 | Joback Method |
| dvisc | 0.0000040 | Pa×s    | 652.09 | Joback Method |

## Sources

|                        |                                                                                                                                               |
|------------------------|-----------------------------------------------------------------------------------------------------------------------------------------------|
| <b>NIST Webbook:</b>   | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C10139841&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C10139841&amp;Units=SI</a> |
| <b>Crippen Method:</b> | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>                                     |
| <b>Crippen Method:</b> | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>                             |
| <b>Joback Method:</b>  | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                         |
| <b>McGowan Method:</b> | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                         |

## Legend

|                 |                                                 |
|-----------------|-------------------------------------------------|
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>dvisc:</b>   | Dynamic viscosity                               |
| <b>gf:</b>      | Standard Gibbs free energy of formation         |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mcvol:</b>   | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>ripol:</b>   | Polar retention indices                         |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tc:</b>      | Critical Temperature                            |
| <b>tf:</b>      | Normal melting (fusion) point                   |
| <b>vc:</b>      | Critical Volume                                 |

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