

# Hexanoic acid, 2-methoxyethyl ester

<b>Other names:</b>	2-Methoxyethyl caproate
<b>Inchi:</b>	InChI=1S/C9H18O3/c1-3-4-5-6-9(10)12-8-7-11-2/h3-8H2,1-2H3
<b>InchiKey:</b>	WHFYLHIOEPSQJH-UHFFFAOYSA-N
<b>Formula:</b>	C9H18O3
<b>SMILES:</b>	CCCCCC(=O)OCCOC
<b>Mol. weight [g/mol]:</b>	174.24

## Physical Properties

Property code	Value	Unit	Source
gf	-314.02	kJ/mol	Joback Method
hf	-606.11	kJ/mol	Joback Method
hfus	23.04	kJ/mol	Joback Method
hvap	47.19	kJ/mol	Joback Method
log10ws	-1.54		Crippen Method
logp	1.756		Crippen Method
mvol	150.980	ml/mol	McGowan Method
pc	2372.59	kPa	Joback Method
rinpol	1183.00		NIST Webbook
rinpol	1183.00		NIST Webbook
tb	504.03	K	Joback Method
tc	677.20	K	Joback Method
tf	285.58	K	Joback Method
vc	0.582	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	348.03	J/mol×K	504.03	Joback Method
cpg	361.09	J/mol×K	532.89	Joback Method
cpg	373.72	J/mol×K	561.75	Joback Method
cpg	385.91	J/mol×K	590.62	Joback Method
cpg	397.66	J/mol×K	619.48	Joback Method
cpg	408.97	J/mol×K	648.34	Joback Method
cpg	419.82	J/mol×K	677.20	Joback Method

dvisc	0.0024200	Paxs	285.58	Joback Method
dvisc	0.0012601	Paxs	321.99	Joback Method
dvisc	0.0007491	Paxs	358.40	Joback Method
dvisc	0.0004902	Paxs	394.81	Joback Method
dvisc	0.0003446	Paxs	431.21	Joback Method
dvisc	0.0002559	Paxs	467.62	Joback Method
dvisc	0.0001984	Paxs	504.03	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U330926&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U330926&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/81-479-1/Hexanoic-acid-2-methoxyethyl-ester.pdf>

Generated by Cheméo on 2024-04-26 05:52:31.23172288 +0000 UTC m=+16400000.152300190.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.