

1,3-Propanediol, O-chlorodifluoroacetate-O'-heptafluorobutyrate-

Inchi:	InChI=1S/C9H6ClF9O4/c10-7(13,14)5(21)23-3-1-2-22-4(20)6(11,12)8(15,16)9(17,18)19/
InchiKey:	INOCPNWOMFYNAG-UHFFFAOYSA-N
Formula:	C9H6ClF9O4
SMILES:	O=C(OCCCC(=O)C(F)(F)C(F)(F)C(F)(F)F)C(F)(F)Cl
Mol. weight [g/mol]:	384.58

Physical Properties

Property code	Value	Unit	Source
gf	-2196.80	kJ/mol	Joback Method
hf	-2534.42	kJ/mol	Joback Method
hfus	26.90	kJ/mol	Joback Method
hvap	45.79	kJ/mol	Joback Method
log10ws	-3.57		Crippen Method
logp	3.127		Crippen Method
mvol	180.720	ml/mol	McGowan Method
pc	1801.56	kPa	Joback Method
rinpol	1060.00		NIST Webbook
rinpol	1060.00		NIST Webbook
tb	575.84	K	Joback Method
tc	733.77	K	Joback Method
tf	380.42	K	Joback Method
vc	0.754	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	482.75	J/mol×K	575.84	Joback Method
cpg	492.49	J/mol×K	602.16	Joback Method
cpg	501.55	J/mol×K	628.48	Joback Method
cpg	509.96	J/mol×K	654.81	Joback Method
cpg	517.76	J/mol×K	681.13	Joback Method
cpg	524.99	J/mol×K	707.45	Joback Method
cpg	531.67	J/mol×K	733.77	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U375557&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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